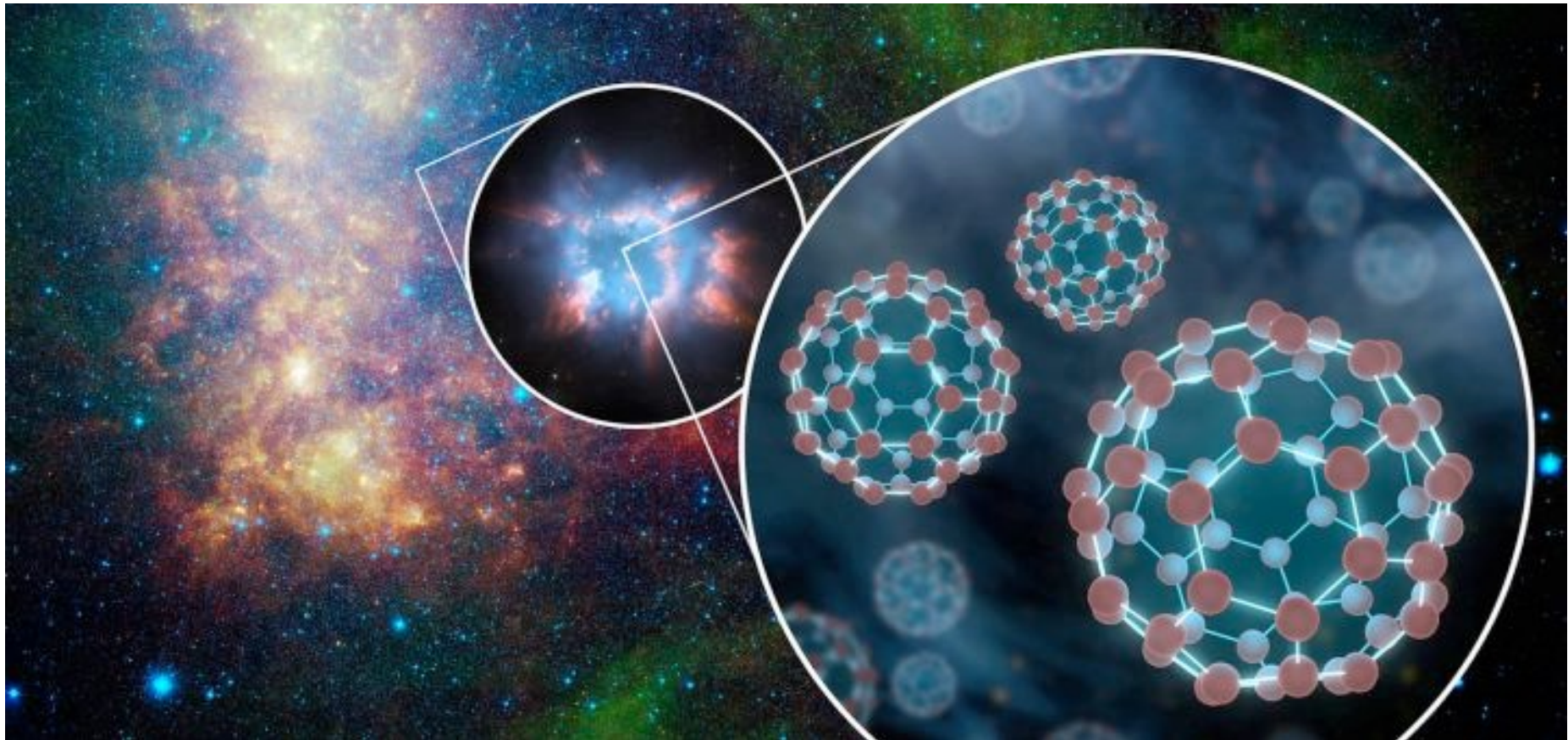


# Chemistry of interstellar clouds

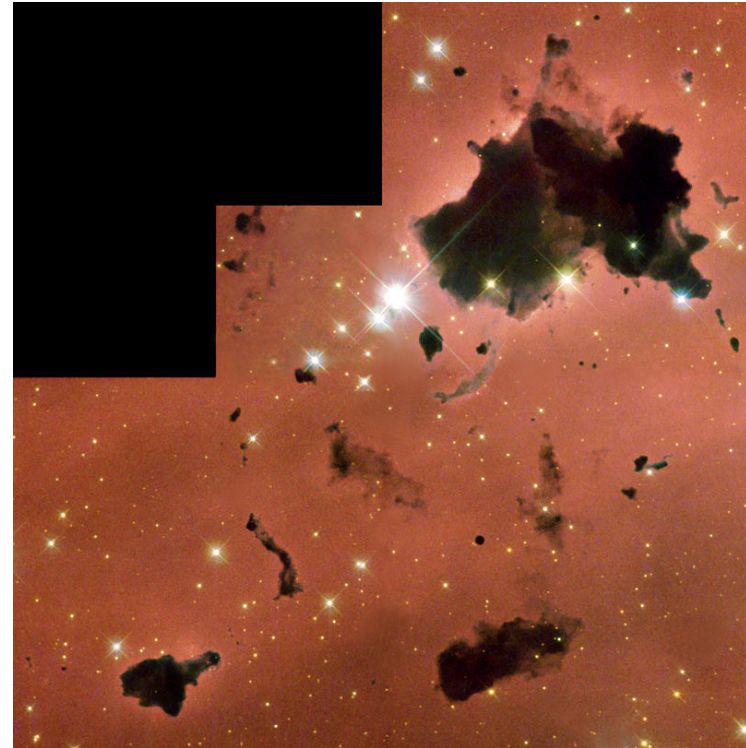
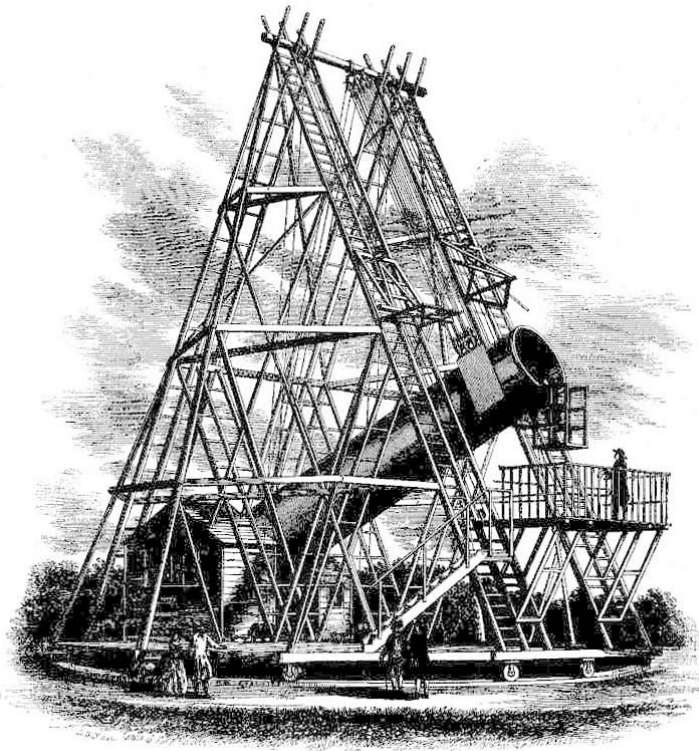
Floris van der Tak



# The sky: more than just stars



# Telescopes: Holes in the sky





# Interstellar gas clouds & molecules

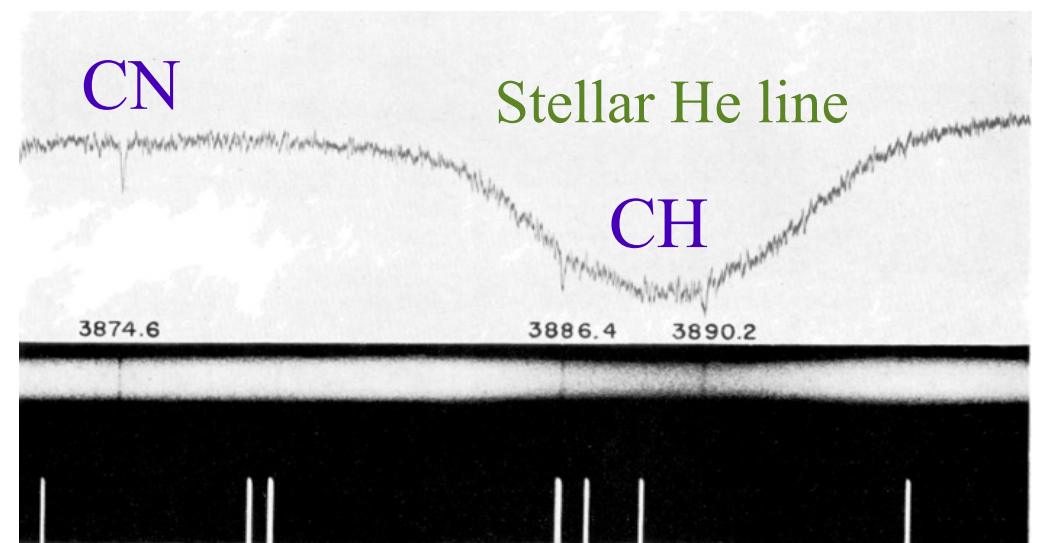
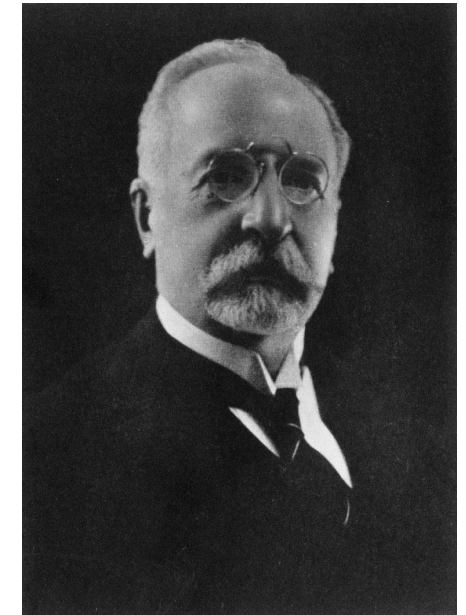
Hartmann 1904: stationary Na absorption  
in spectrum of ‘spectroscopic binary’

Eddington 1926: thin ionized ISM,  $10^4$ – $10^6$  K  
describes *intergalactic* medium

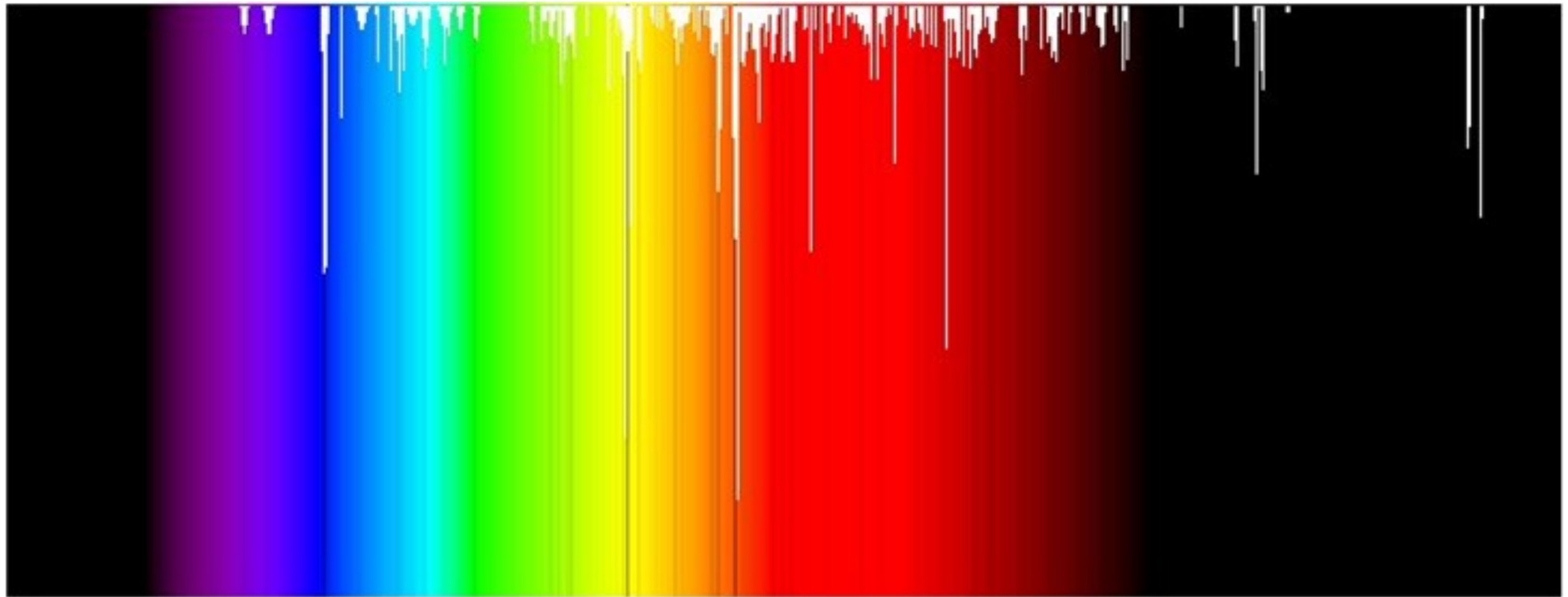
~1940: First gas-phase molecules  
CH, CN, CH<sup>+</sup>

Known from comets since ~1870  
Puzzle: low excitation ( $T_{\text{ex}} = 3$  K)

*Spectrum of  $\zeta$  Oph (1941)*



# The ‘diffuse’ interstellar bands: A puzzle for >100 years



Over 300 broad absorption features known from UV to near-IR

also in ‘Magellanic Clouds’ = Milky Way satellites

Remain mostly unidentified since 1922!

Must be exceptionally stable: survive harsh conditions

probably medium-large hydrocarbons (Linnartz +2010, Oka +2013)

Most plausible claim so far:  $C_{60}^+$  (Campbell / Maier 2015-2018, 5 features)

# Beyond optical astronomy

- **Radio astronomy (1950s-1960s)**

- [H I] 21 cm: weak hyperfine transition

- massive cold interstellar gas clouds

- OH 18 cm: maser *emission*

- collisional pumping = high gas density

- NH<sub>3</sub>, H<sub>2</sub>O, H<sub>2</sub>CO: polyatomics

- significant chemistry

- **High energy astrophysics (1970s)**

- UV spectra: H<sub>2</sub> as abundant as H

- shielding; importance grain surface chemistry

- X-ray spectra: atomic composition of dust

- elemental depletion in ISM wrt Sun

- **Far-IR & submillimeter (1980s)**

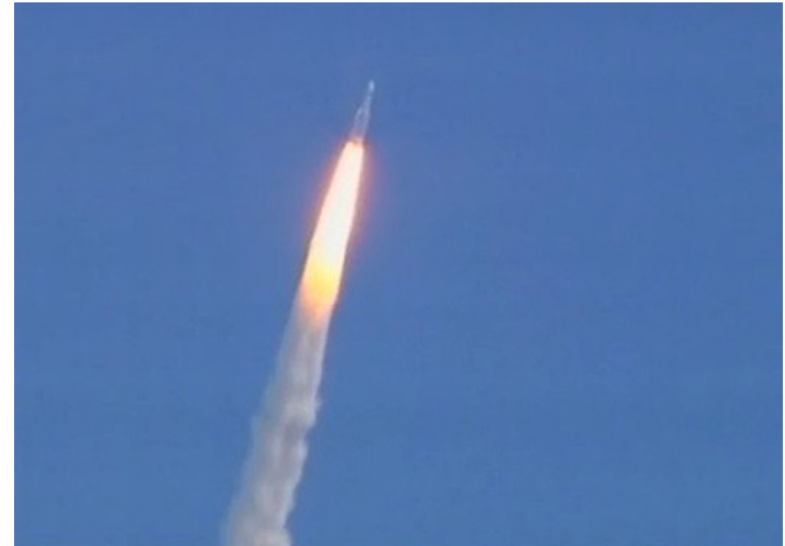
- CO: important to trace cold (=bulk) H<sub>2</sub>

- hydride: large level spacing

- (lack of dipole moment is advantage!)

- X-ogen = HCO<sup>+</sup>: space < lab

- prevalence ions, radicals, C-chains

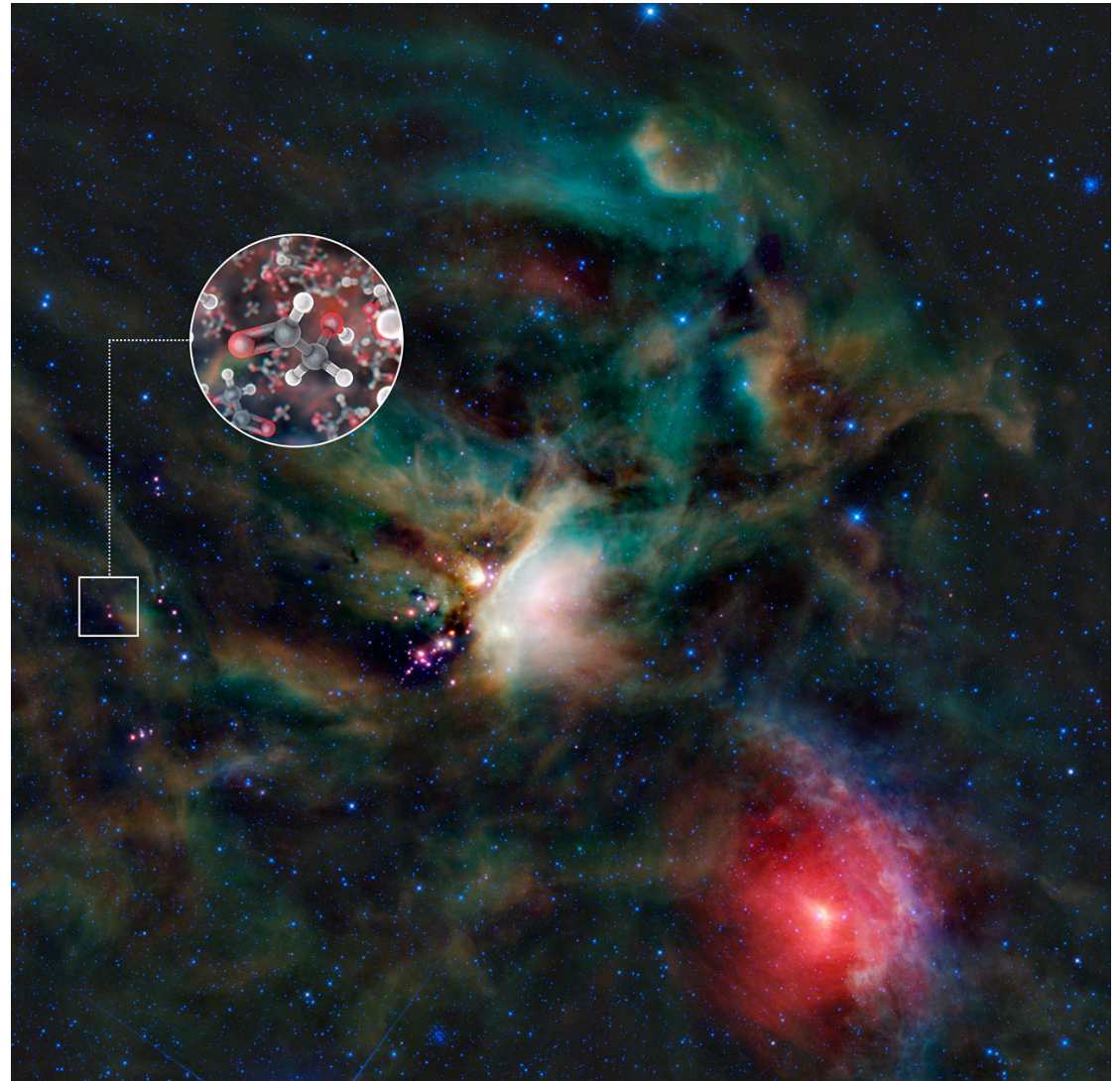


# Today's workhorse #1: ALMA

Glycoaldehyde around a young Solar-type star

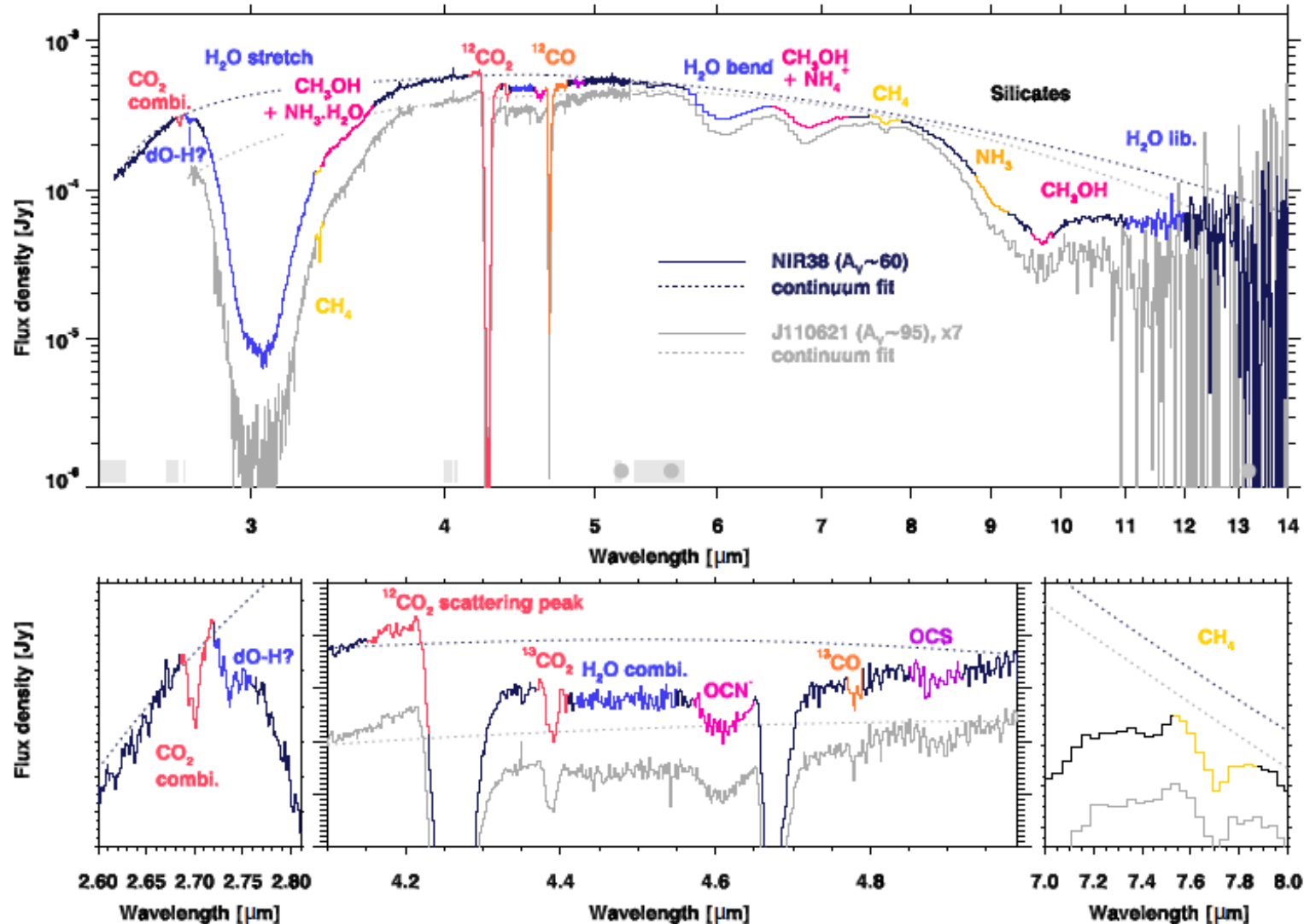
Building blocks for life exist in right time at right place to be included in planets

Formation in 'heated' ice layer?





# Today's workhorse #2: JWST



Ice mantles due to condensation of volatile species onto dust grains  
visible as broad absorption features in mid-IR  
*JWST spectra of interstellar clouds: McClure +2023 Nature Astronomy*

# New molecular detections 2023/2024

FeC	KP 12m	IRC 10216	(HO) <sub>2</sub> CO	Yebes	IRC 10216
HSO	Yebes	Dense cores	NC <sub>4</sub> NH <sup>+</sup>	Yebes	IRC 10216
HCNS	IRAM 30m	TMC-1	C <sub>7</sub> N <sup>-</sup>	Yebes	TMC-1, IRC 10216
NaC <sub>3</sub> N	Yebes	IRC 10216	C <sub>10</sub> H <sup>-</sup>	GBT	TMC-1, IRC 10216
MgC <sub>3</sub> N <sup>+</sup> , MgC <sub>5</sub> N <sup>+</sup>	Yebes	IRC 10216	CH <sub>3</sub> CHCO	Yebes	TMC-1
HMgC <sub>3</sub> N	Yebes	IRC 10216	HOCH <sub>2</sub> C(O)NH <sub>2</sub>	IRAM 30m	G0.693
MgC <sub>4</sub> H <sup>+</sup> , MgC <sub>6</sub> H <sup>+</sup>	Yebes	IRC 10216	H <sub>2</sub> C(CH) <sub>3</sub> CN	GBT	TMC-1
H <sub>2</sub> C <sub>3</sub> H <sup>+</sup>	Yebes	IRC 10216	(CH <sub>3</sub> ) <sub>2</sub> C=CH <sub>2</sub>	IRAM 30m	G0.693
H <sub>2</sub> C <sub>3</sub> N	Yebes	IRC 10216	C <sub>6</sub> CH <sub>5</sub> C <sub>2</sub> H	Yebes	TMC-1

*Some anions*

*Glycine isomer*

*Some cations*

# This lecture

- Motivation & context
- **Physics of the interstellar medium**
  - atoms
  - radiation
  - dust grains
  - cosmic rays
  - clouds
- **Basic interstellar chemistry**
  - gas-phase processes
  - grain surface processes
  - diffuse clouds
  - dark / dense clouds









# The building blocks of molecules

**Periodic Table of Elements**

1A	1	H	IIA	2	He	0																			
	2	Li	Be		5	B	6	C	7	N	8	O	9	F	10	Ne									
	3	Na	Mg	III B	IV B	V B	VI B	VII B	VIII	IX	X	IB	II B	13	Al	14	Si	15	P	16	S	17	Cl	18	Ar
	4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	31	Ga	32	Ge	33	As	34	Se	35	Br	36	Kr
	5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	49	In	50	Sn	51	Sb	52	Te	53	I	54	Xe
	6	Cs	Ba	*La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	81	Tl	82	Pb	83	Bi	84	Po	85	At	86	Rn
	7	Fr	Ra	+Ac	Rf	Ha	106	107	108	109	110														

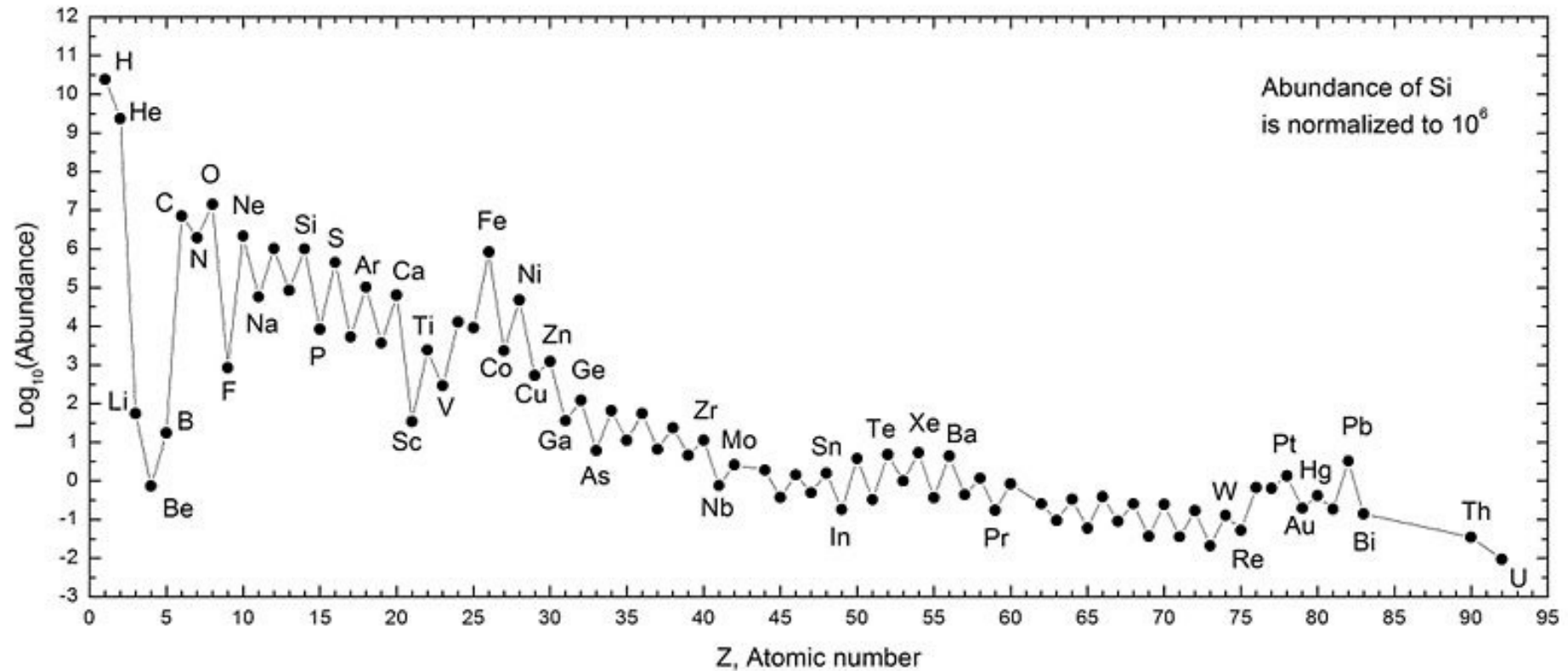
* Lanthanide Series	58	59	60	61	62	63	64	65	66	67	68	69	70	71
	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
+ Actinide Series	90	91	92	93	94	95	96	97	98	99	100	101	102	103
	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

Legend - click to find out more...

<b>H - gas</b>	<b>Li - solid</b>	<b>Br - liquid</b>	<b>Tc - synthetic</b>
 Non-Metals	 Transition Metals	 Rare Earth Metals	 Halogens
 Alkali Metals	 Alkali Earth Metals	 Other Metals	 Inert Elements

*Useful to understand atomic structure & behaviour, but less representative for ISM*

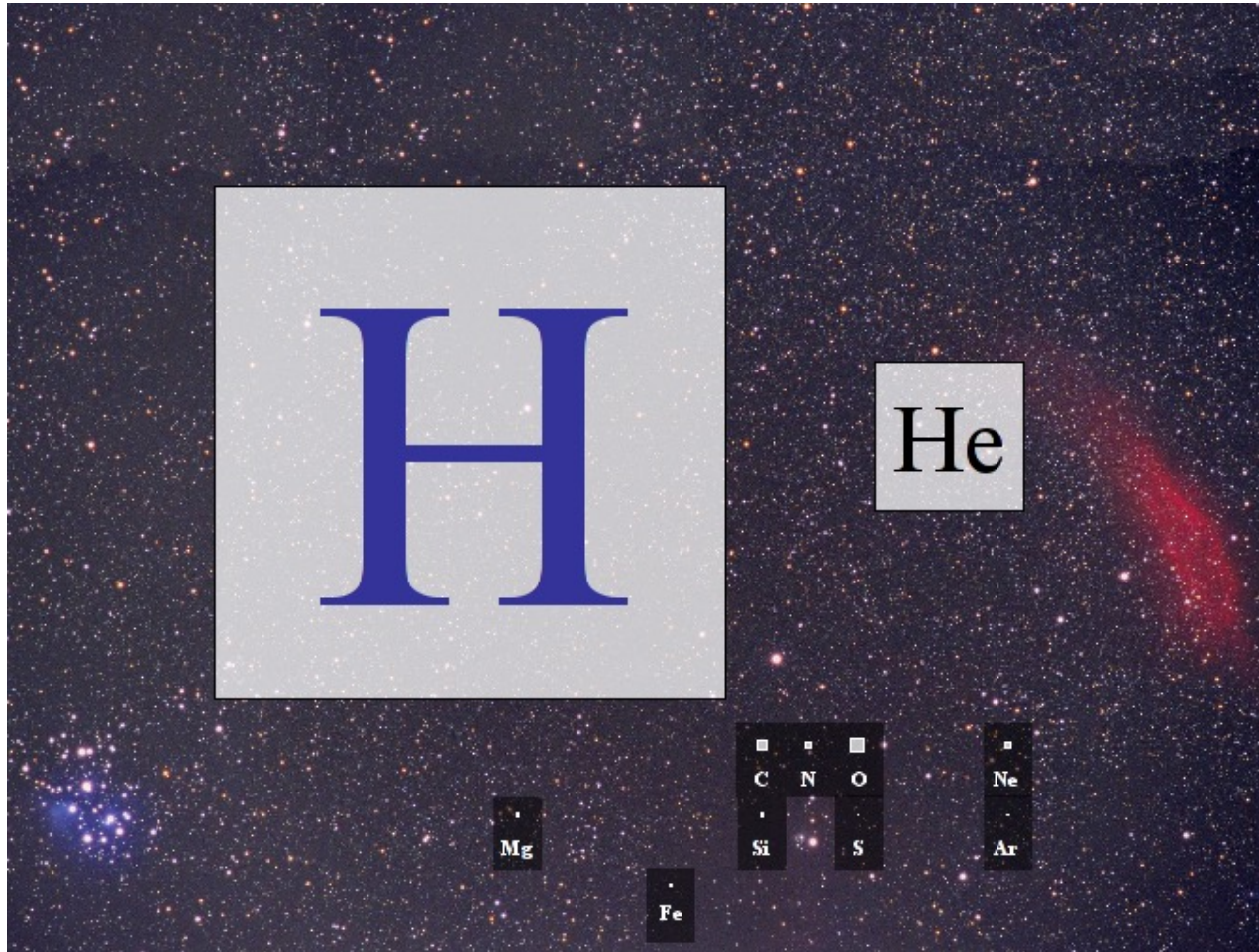
# Cosmic abundances reflect young Universe



## Combined effect of nuclear and stellar physics:

- primordial H, He production
- Li, Be, B destroyed in stars
- dominance of H/He/C burning: odd-even effect +  $\alpha$  capture
- Fe peak: max binding energy
- beyond Fe: peaks around “magic” nuclei (closed neutron shell)

# Astronomers' Periodic Table (Ben McCall)

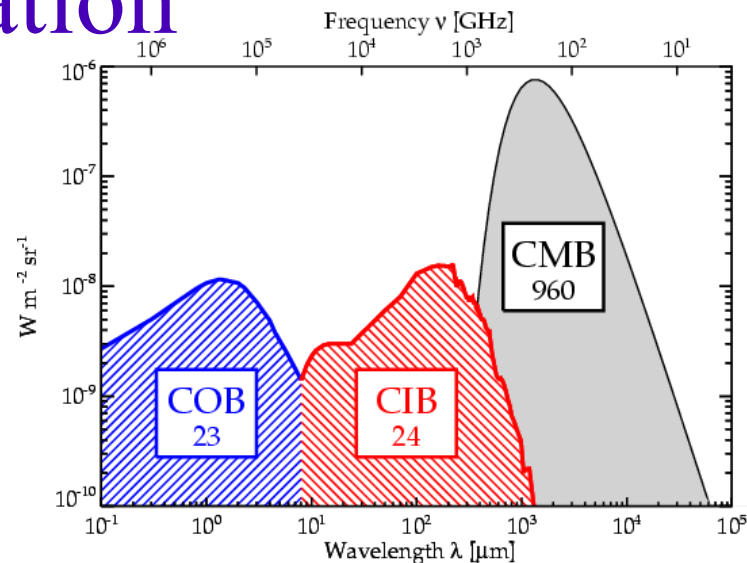


He  $\sim 0.1$ ; C, N, O  $\sim 10^{-4}$ ; Mg, Fe, Si, S  $\sim 10^{-5}$

Expect H-dominated chemistry, C+N+O sidekicks, Si+S bystanders

# Interstellar radiation

- **Components: starlight, dust, and CMB**  
 $\mu$ wave dominates by photons (early Universe)  
infrared: rotational / vibrational excitation  
**vis/UV: ionization & dissociation**



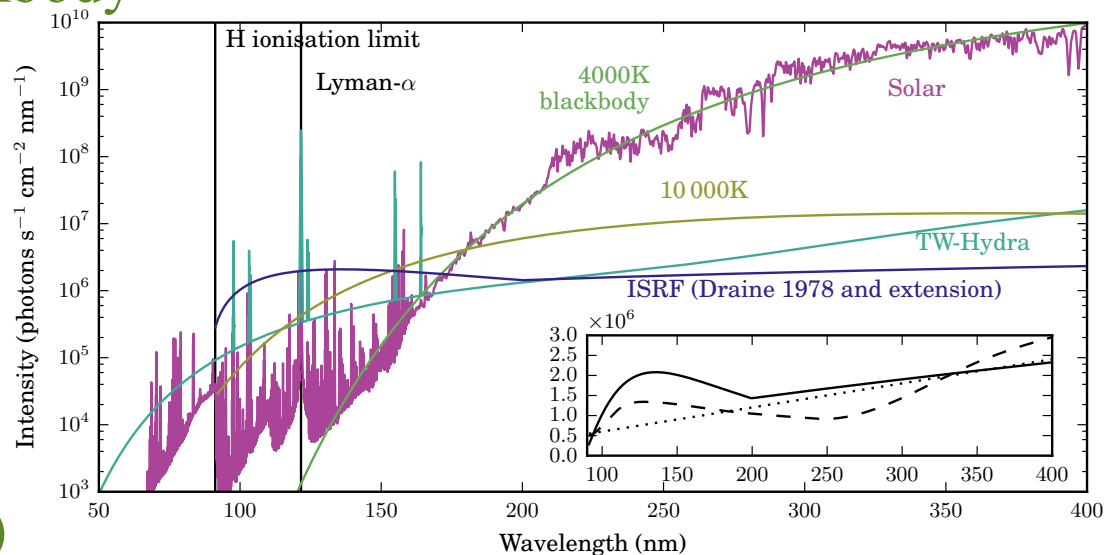
- **UV: Habing (1968) & Draine (1978) estimates**  
agree within  $\sim 2\times$  ( $\chi = 1.7 G_0$ )  
i.e. within variation ( $\sim 3\times$ ) within Galactic plane on stellar ( $\sim$ Gyr) timescale

*Dole et al 2006*

- **Dominated by “B-type” stars ( $\sim 10 M_\odot$ )**  
hot: spectrum like 20,000 K blackbody  
rare: flux  $1.6 \cdot 10^{-3} \text{ erg cm}^{-2} \text{ s}^{-1}$

- **Deviates from blackbody shape**  
cool stars: excess in optical  
atomic hydrogen: 912 Å cutoff

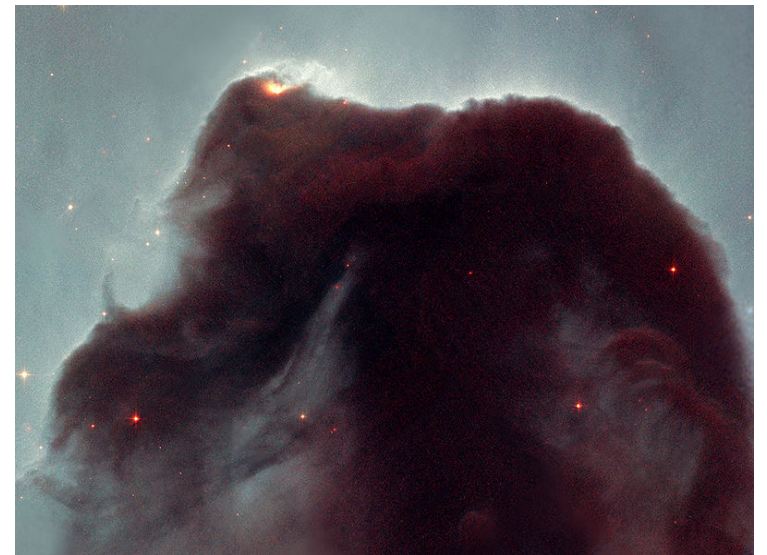
- **Consequences for atomic states**  
neutral if IP  $> 13.6 \text{ eV}$  (He, O, N)  
otherwise ionized (C, Na, Si)



*Heays et al 2017*

# Interstellar dust

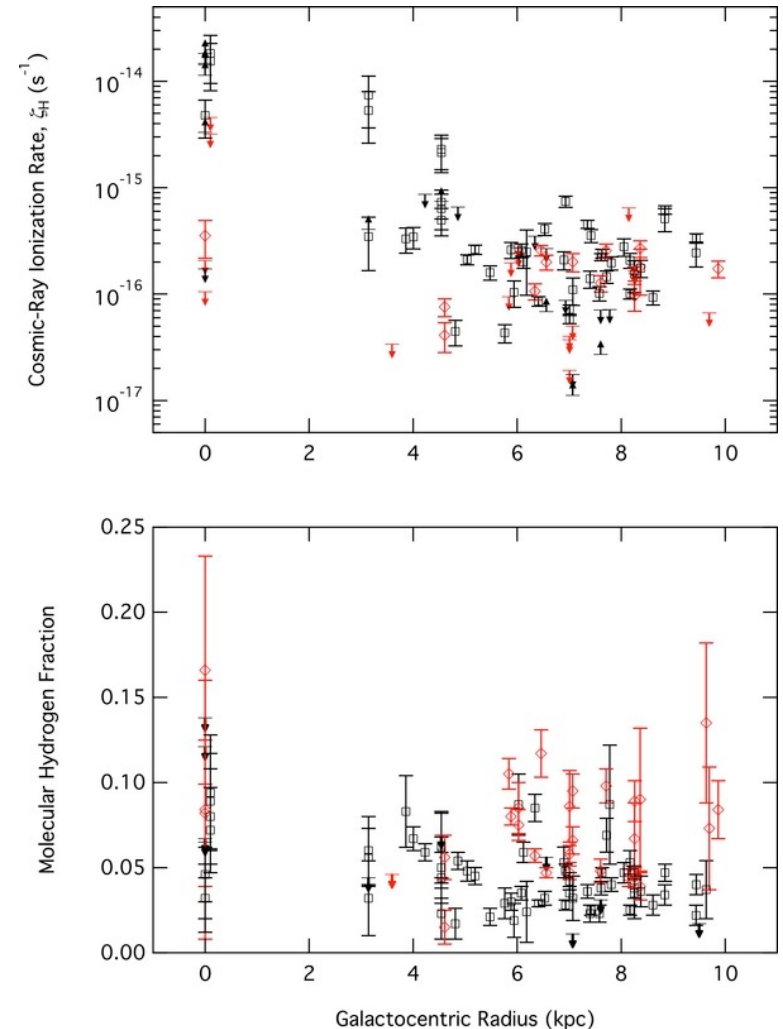
- **Optical extinction, polarization, ‘reddening’**: size, shape, composition  
infrared emission: temperature, mass, ..
- **Solid particles**  
silicate (“sand”) + carbonaceous (“soot”)  
in cold dense clouds: ice mantle
- **Average size 0.1  $\mu\text{m}$**   
range from 0.01 to 0.5  $\mu\text{m}$   
larger in protoplanetary disks
- **Abundance 1% by mass,  $10^{-12}$  by number**  
silicate core contains most Si, Fe, Mg  
carbonaceous part ~60% of C, 30% of O
- **Created in outflows from evolved stars**  
destroyed in shock waves  
consumed in planet formation  
stellar explosions: both form & destroy





# Cosmic rays

- Energetic (MeV – GeV) nuclei traversing the Galaxy accelerated in stellar explosions (“supernovae”) deflected by magnetic fields
- Interact with molecular clouds ionization induced UV field
- Ionization rate  $10^{-17} \dots 10^{-16} \text{ s}^{-1}$  lower in dense clouds higher near Galactic Center influence chemistry & dynamics
- Induced UV field  $\sim 1\%$  of interstellar average photodissociation heating by photo-electrons photodesorption from dust grains



# Interstellar clouds: The birthplaces of stars & planets

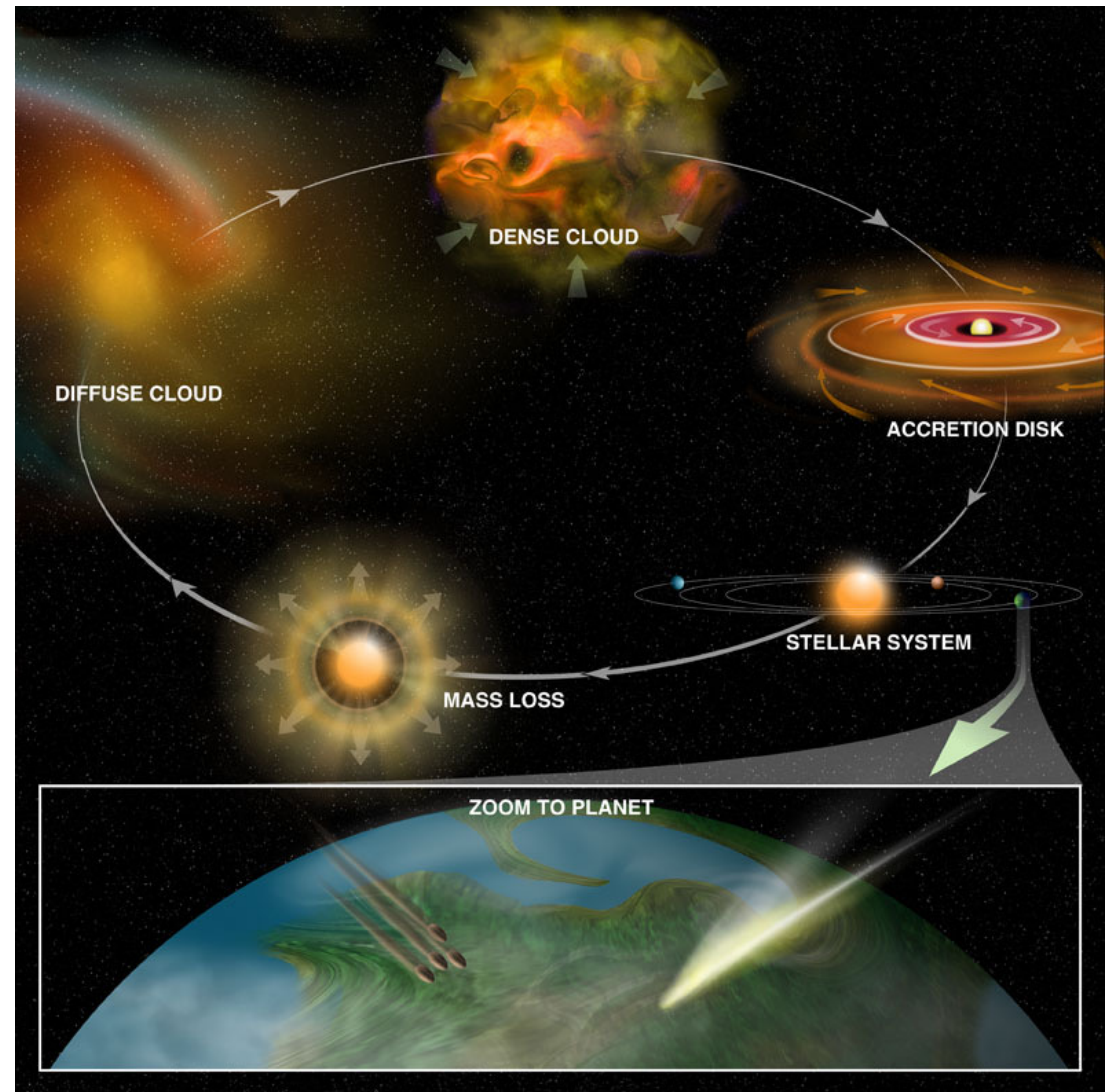
Molecules found so far:

>300 in ISM / CSM  
(+20 since 2023)

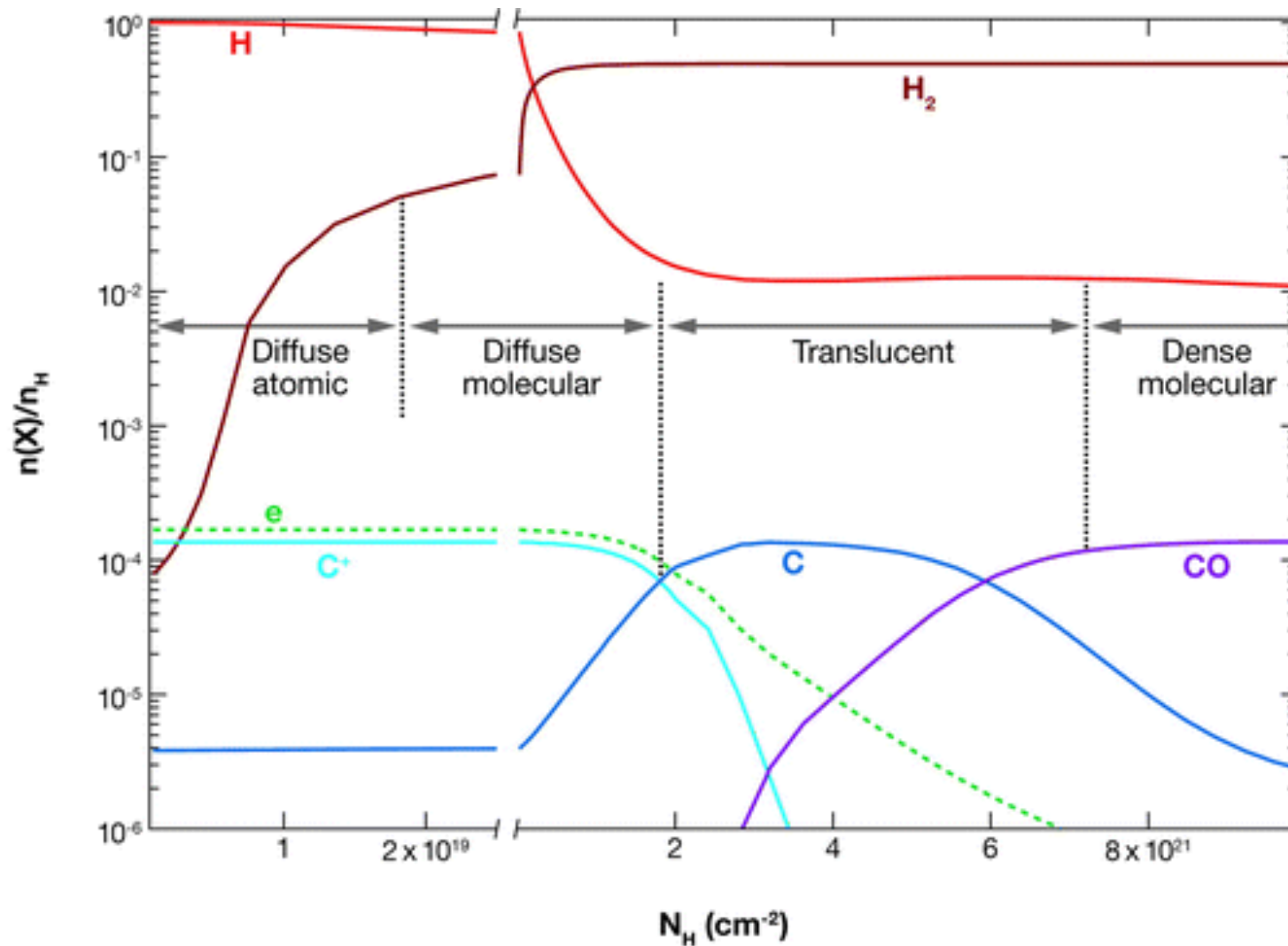
~73 extragalactic


~10 in early Universe

~16 in exoplanet  
atmospheres



# Types of interstellar clouds



 Snow TP, McCall BJ. 2006.  
Annu. Rev. Astron. Astrophys. 44:367–414

Clouds named after dominant form of hydrogen:  
ionized / atomic / molecular  
Ionized clouds dominate ISM by volume,  
molecular clouds by mass

# This lecture

- **Motivation & context**
- **Physics of the interstellar medium**
  - atoms
  - radiation
  - dust grains
  - cosmic rays
  - clouds
- **Basic interstellar chemistry**
  - gas-phase processes
  - grain surface processes
  - diffuse clouds
  - dark / dense clouds

# Basic interstellar chemistry: Gas phase

- **Low temperatures and densities**  
chemistry far from thermodynamic equilibrium  
reactions with significant activation barriers inhibited
- **Limited by kinetics; only two-body reactions**  
3-body interactions enter at  $n > 10^{12} \text{ cm}^{-3}$  (e.g. inner disks)
- **Hydrogen dominant element**  
reactions with H & H<sub>2</sub> preferred if exoergic
- **Time dependence often important**  
other parameters:  $T$ ,  $n$ ,  $Z$ , radiation, ...
- **Modern models: 1000s of reactions**  
of just 9 different types

# The 3x3 types of chemical reactions

<b>Formation of bonds</b>		
radiative association	$X + Y^+ \rightarrow XY^+ + h\nu$	Diffuse clouds
associative detachment	$X + Y^- \rightarrow XY + e$	Diffuse clouds
grain surface	$X + Y:g \rightarrow XY + g$	Dense clouds

# The 3x3 types of chemical reactions

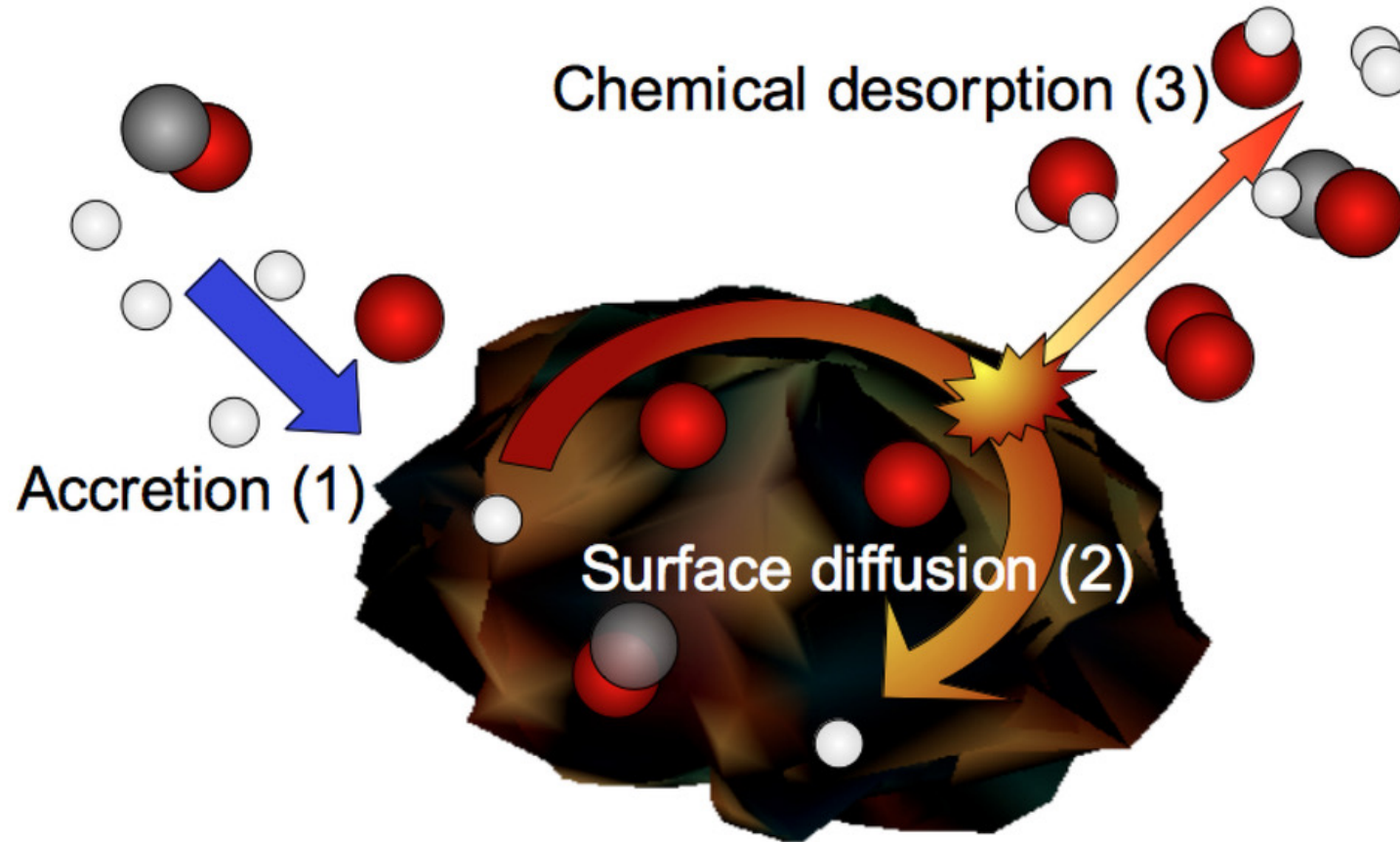
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associative detachment	$X + Y^- \rightarrow XY + e$	Diffuse clouds
grain surface	$X + Y:g \rightarrow XY + g$	Dense clouds
<b>Destruction of bonds</b>		
photodissociation	$XY + h\nu \rightarrow X + Y$	Stellar vicinities
dissociative recombination	$XY^+ + e \rightarrow X + Y$	Diffuse clouds
collisional dissociation	$XY + M \rightarrow X + Y + M$	Shocks

# The 3x3 types of chemical reactions

<b>Formation of bonds</b>		
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dissociative recombination	$XY^+ + e \rightarrow X + Y$	Diffuse clouds
collisional dissociation	$XY + M \rightarrow X + Y + M$	Shocks
<b>Rearrangement of bonds</b>		
ion-molecule reaction	$X^+ + YZ \rightarrow XY^+ + Z$	Cold dense clouds
charge transfer reaction	$X^+ + YZ \rightarrow X + YZ^+$	Diffuse clouds
neutral-neutral reaction	$X + YZ \rightarrow XY + Z$	Warm dense clouds

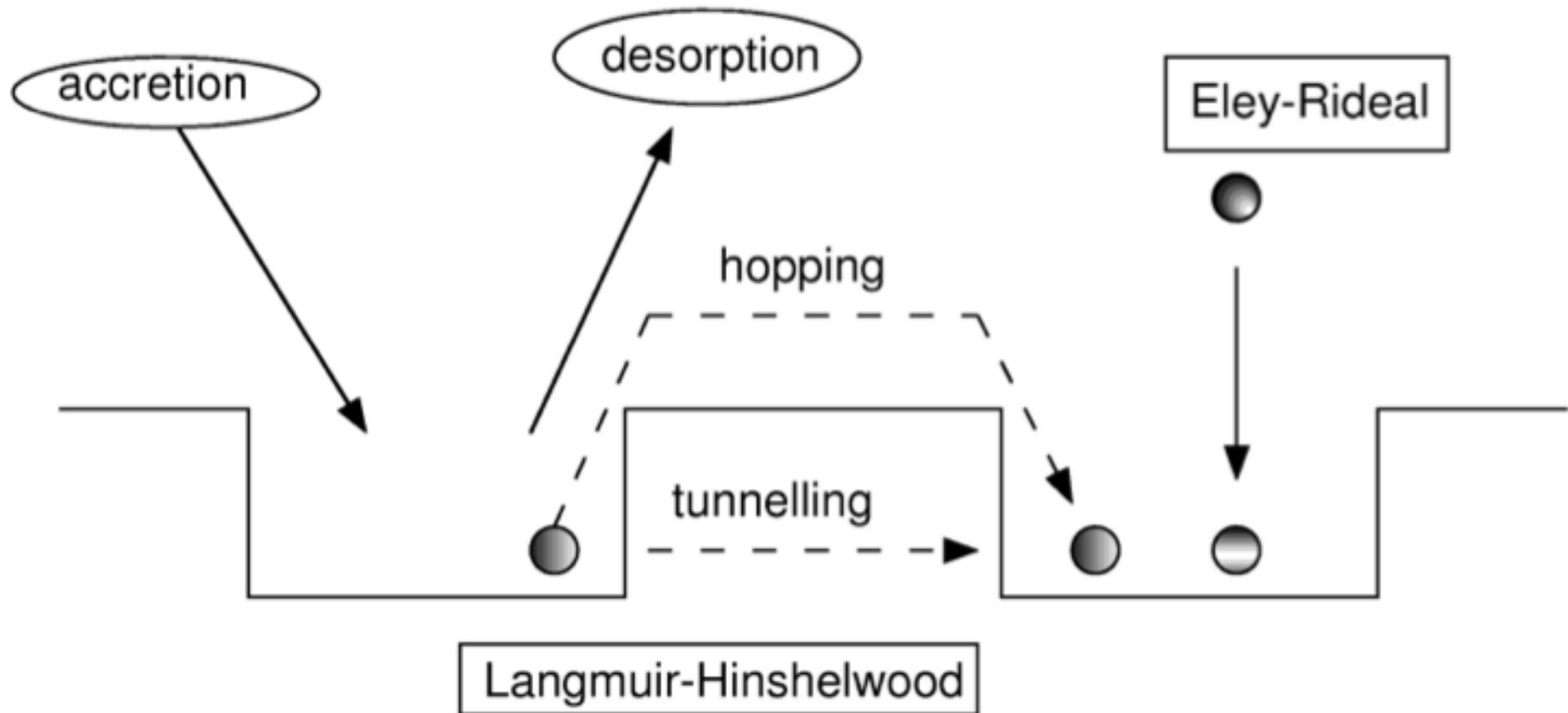


# Basic processes: Grain surface chemistry



evidence: interstellar  $\text{H}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{CH}_3\text{OH}$ ;  $\text{CO}_2$  ice, ...

# Mechanisms of grain surface chemistry



LH route: sticking / diffusion / reaction / desorption

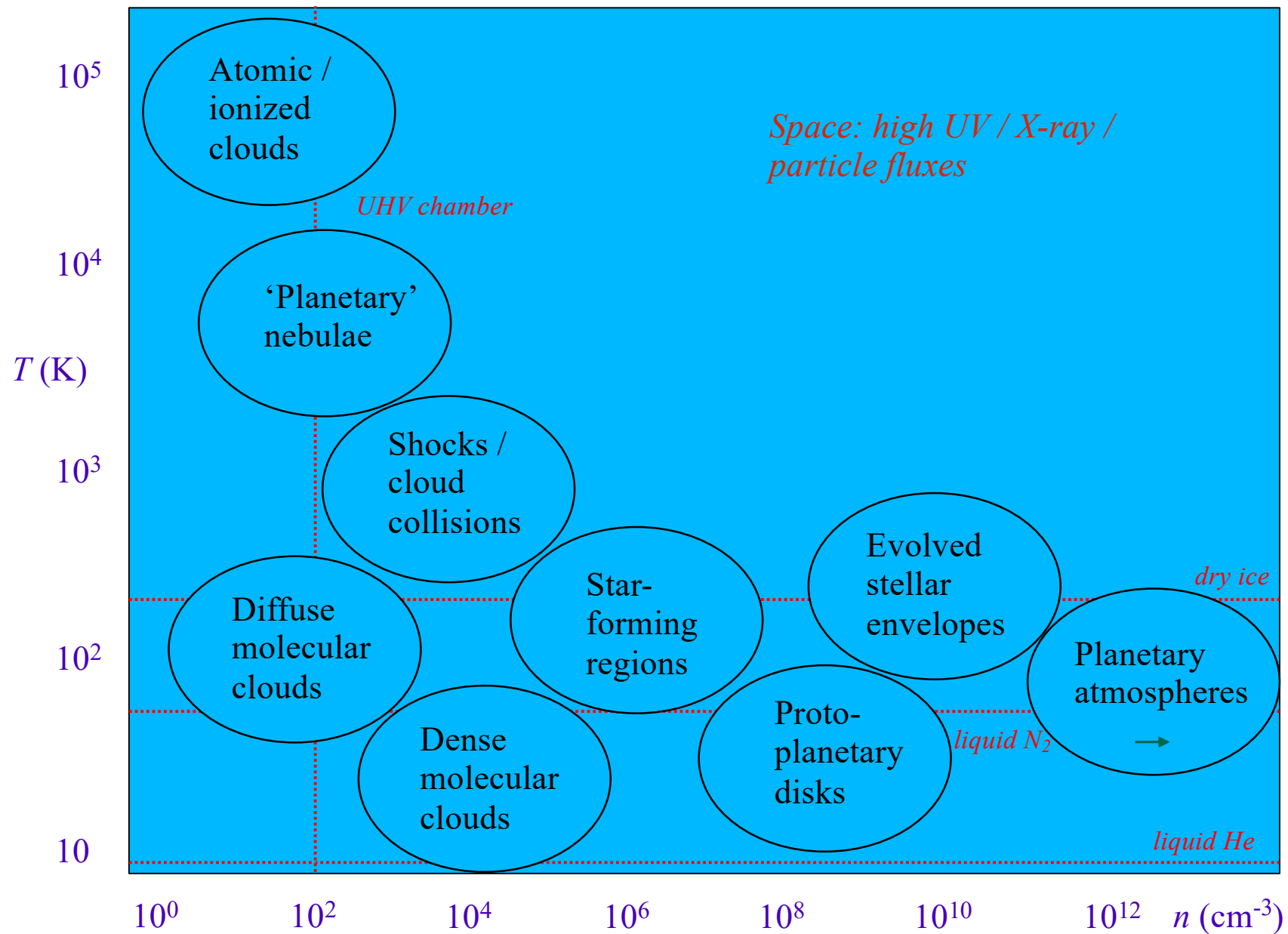
ER route: direct reaction + desorption

Grain surface can be silicate, carbonaceous, or ice

Reaction rates hard to measure (unlike gas phase)

Modern models include radical reactions within heated ice layer

# Interstellar space: Wide range of conditions



# Observations of diffuse clouds

Conditions:  $T = 100$  K,  $n = 10^2 - 10^3$  cm $^{-3}$

atomic lines: depletion of C, N, O, ...  
chemistry sensitive to C/O ratio

- **Strong VIS/UV absorption lines**

H<sub>2</sub>, HD (ground / FUSE / HST)  
CH, CH<sup>+</sup>, C<sub>2</sub>, C<sub>3</sub>  
OH, OH<sup>+</sup>, CO  
NH, CN, HCl

- **Weak mm-wave emission lines**

CO, HCO<sup>+</sup>, CN, CS

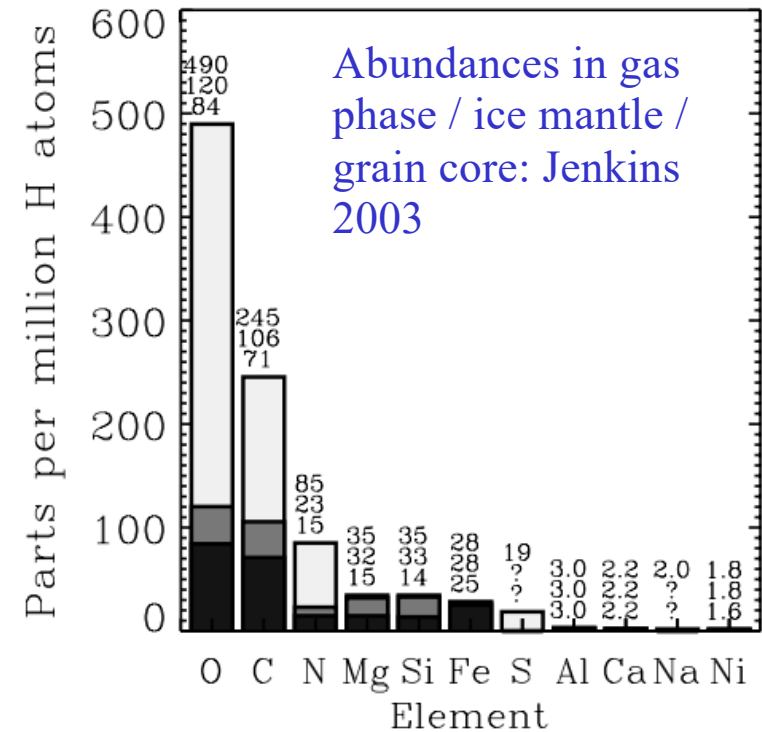
... and absorption: CO, HCO<sup>+</sup>, HCN, ... (Lucas & Liszt 1995–2014)

- **Herschel & SOFIA: added hydrides** (Gerin et al 2016)

H<sub>2</sub>O, H<sub>2</sub>O<sup>+</sup>, H<sub>3</sub>O<sup>+</sup>, NH<sub>2</sub>, NH<sub>3</sub>, SH, SH<sup>+</sup>, ArH<sup>+</sup>, ...

- **ALMA: added organics up to N=7**

H<sub>2</sub>CO, CH<sub>3</sub>CN, c-C<sub>3</sub>H<sub>2</sub>, HC<sub>3</sub>N



# Chemical networks for diffuse clouds

Carbon chemistry starts from RA of  $C^+$  with  $H_2$  into  $CH_2^+ + hv$

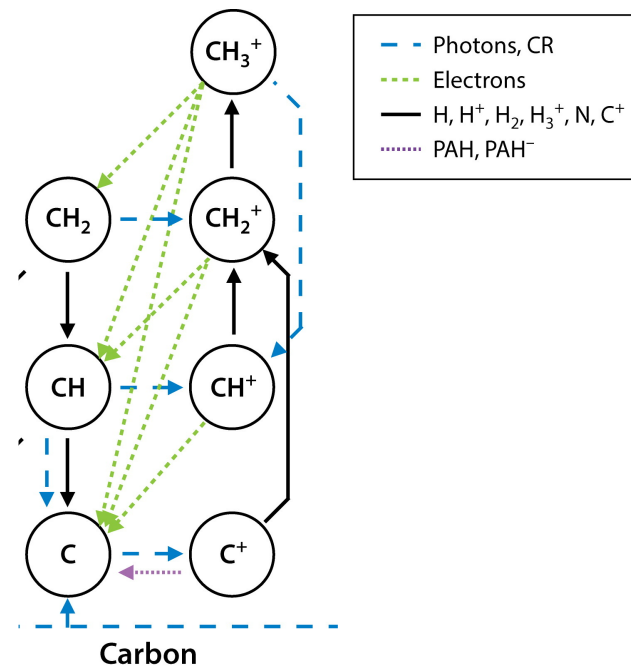
ion-molecule reactions: small (ionized) hydrides

Oxygen chemistry needs CR ionization of  $H_2$  & charge transfer to O

mainly OH &  $H_2O$ ; CO acts as sink

Nitrogen like oxygen; both connected to carbon via  $C^+$  reaction

note absence direct N-O connection



# Example: CH

- Formation mostly by RA of C<sup>+</sup> and H<sub>2</sub>  
destruction mostly by photodissociation

$$n(\text{CH}) \propto \frac{n(\text{C}^+)n(\text{H}_2)k_{\text{RA}}}{k_{\text{pd}}}$$

- Rates known within factor of 2  
possible to fit observed column densities
- Most but not all CH forms by this scheme  
up to 50% forms by other processes
- Goal: model that reproduces *all* species  
with same set of conditions / parameters

# Depth-dependent models

UV field is attenuated

photodissociation rate decreases with depth  
realistic models must be depth-dependent

Most reactions are fast (except H<sub>2</sub> formation)

steady state = good assumption

$$\frac{dn_i(z)}{dt} = F_i - n_i D_i$$

=> density

$$n_i(z) = \frac{F_i}{D_i} \text{ cm}^{-3}$$

column density

$$N_i = \int n_i(z) dz \text{ cm}^{-2}$$

*Q: Why do  $F_i$  &  $D_i$  have different dimensions?*

# Rate of CO photodissociation

Line by line calculation

+ CO-H<sub>2</sub> shielding

+ dust attenuation

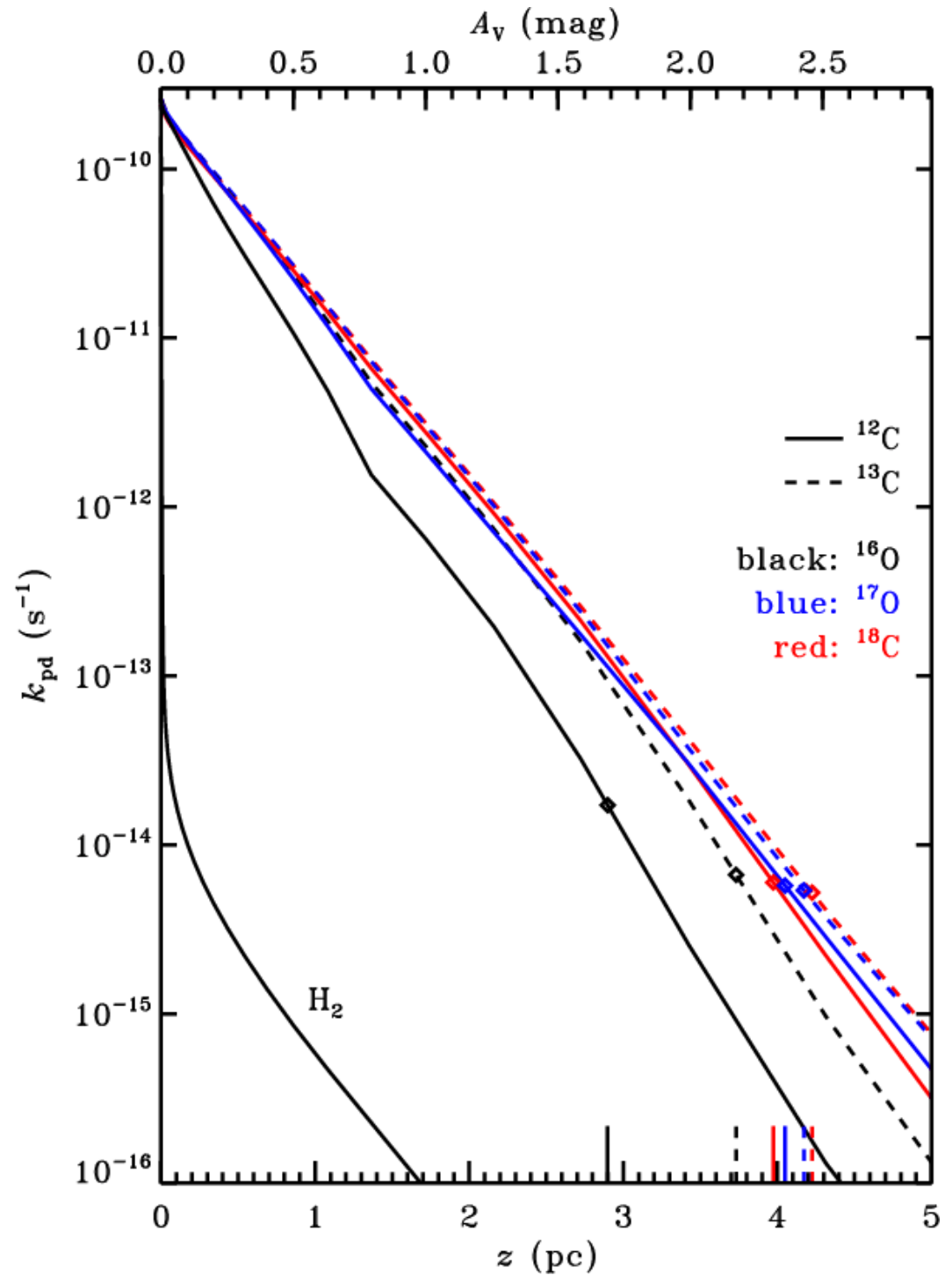
Model parameters:

$$n_{\text{H}} = 300 \text{ cm}^{-3}$$

$$T_{\text{kin}} = 50 \text{ K}$$

$$\chi = 1$$

*Visser et al 2009*



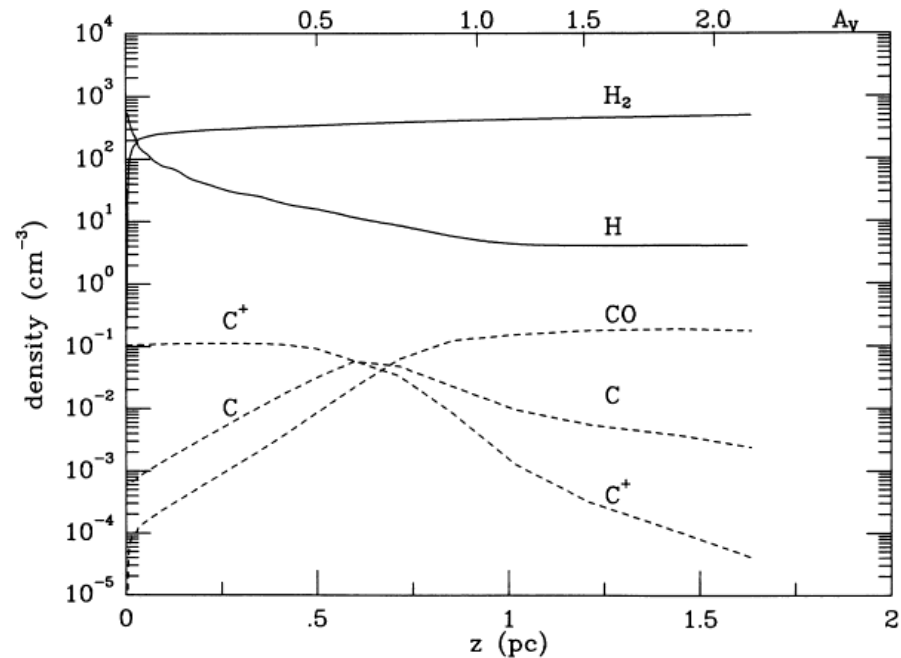


# CO formation and destruction

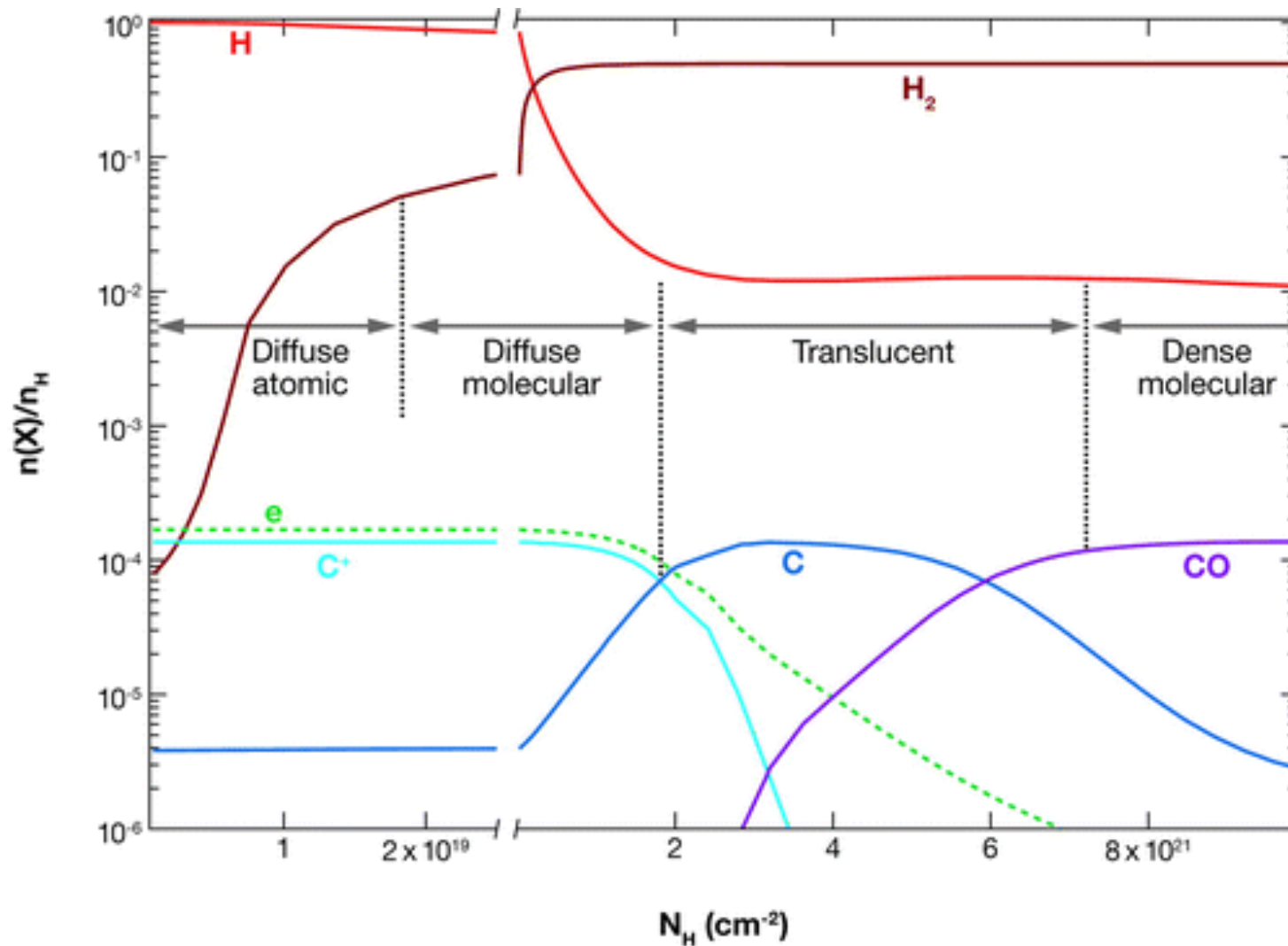
Most abundant molecule after  $H_2$   
easily observed at radio frequencies  
good tracer of cold (=most)  $H_2$


Very stable molecule:  
only dissociated by 912 – 1118 Å photons  
 $D_e = 11.09$  eV  
triple bond

Carbon chemistry:  
at cloud edge: mostly  $C^+$   
interior: mostly CO  
interface layer: C



# Types of interstellar clouds



 Snow TP, McCall BJ. 2006.  
Annu. Rev. Astron. Astrophys. 44:367–414

Diffuse cloud chemistry fairly well understood:  
mainly small radicals & ions  
Main outlier CH<sup>+</sup>:  
role of turbulent gas motions?

# From diffuse to dense clouds

Photoprocesses less important  
just CR-induced UV field

Gas-grain interaction more important  
strong depletion of volatile elements

Long chemical timescales  
dependence on initial conditions

# Dark clouds: mm-wave observations

Conditions:  $T = 10$  K,  $n = 10^4 - 10^5$  cm $^{-3}$   
strong mm-wave low- $J$  line emission

## Prototype: TMC-1

found since 1980s: radicals, C-chains  
since 2000s: anions, saturated chains

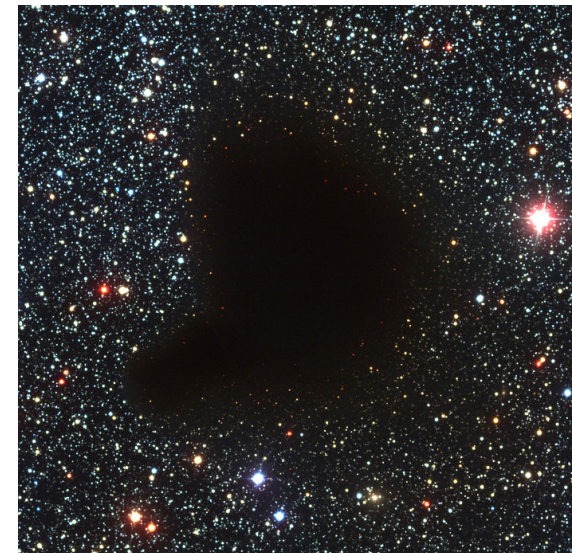
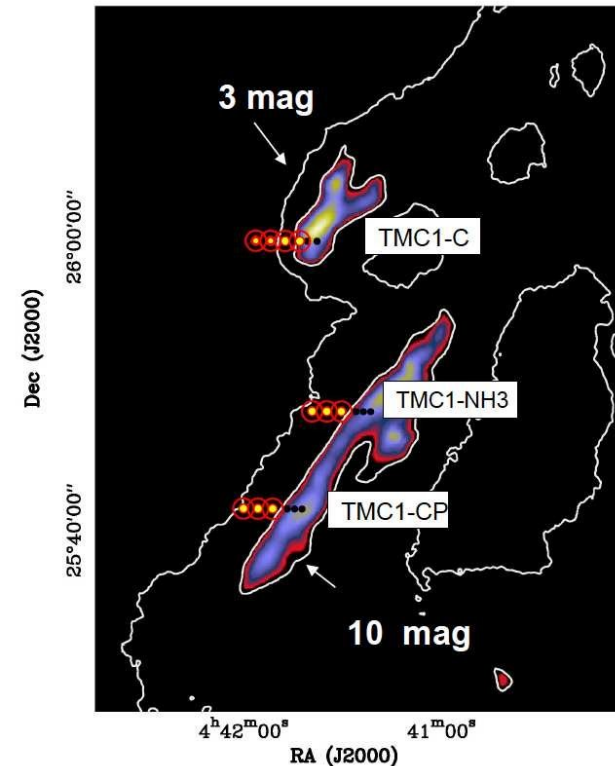
## Abundance variations between cores

NH $_3$  cores: more evolved?  
HC $_x$ N cores: younger?

## Recent focus: pre-stellar cores

prototypes: B68, L1544  
centrally condensed, about to collapse

*Review: Bergin & Tafalla 2007*



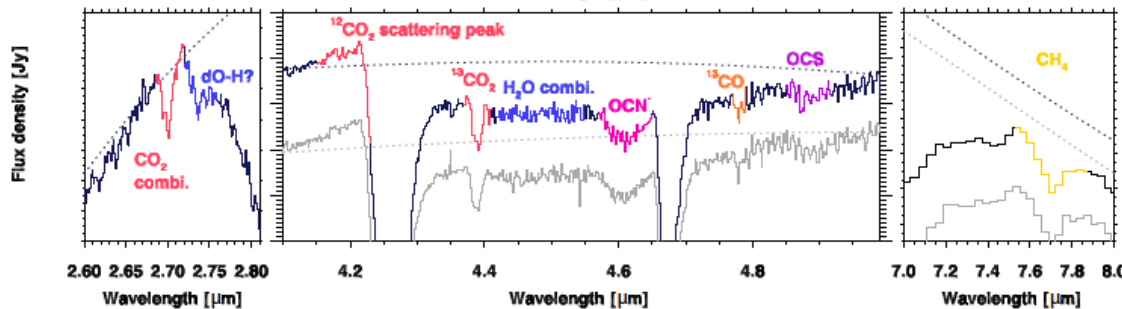
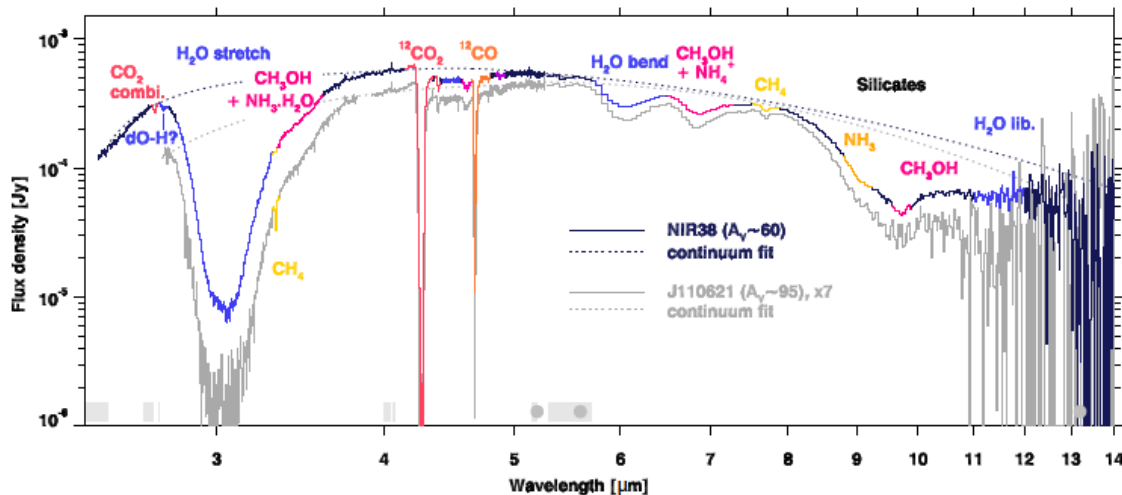
# Mid-IR observations: Solid state

Probe clouds prior to star formation  
unaffected by heating and processing

See same features as toward protostars  
starlight does not change composition

Abundances similar to comets: Sign of inheritance  
Assignment  $\text{NH}_4^+$  disputed, no good alternative

Ice species	Abundance
$\text{H}_2\text{O}$	1
$\text{CO}$	0.27 – 0.32
$\text{CO}_2$	0.19 – 0.35
$\text{CH}_3\text{OH}$	0.02 – 0.05
$\text{NH}_3$	0.05 – 0.10



*IceAge program JWST: McClure +2023*

# Gas-phase models: Time dependence

Most species need  $10^5 - 10^6$  yr to reach equilibrium

solve chemical network as function of time

keep  $T$  &  $n$  constant ( $2 \times 10^4 \text{ cm}^{-3}$ , 10 K)

Initial conditions:

usually  $\text{H}_2$  molecular; C in  $\text{C}^+$ ; He, N, O atomic

represents diffuse clouds

Main result:

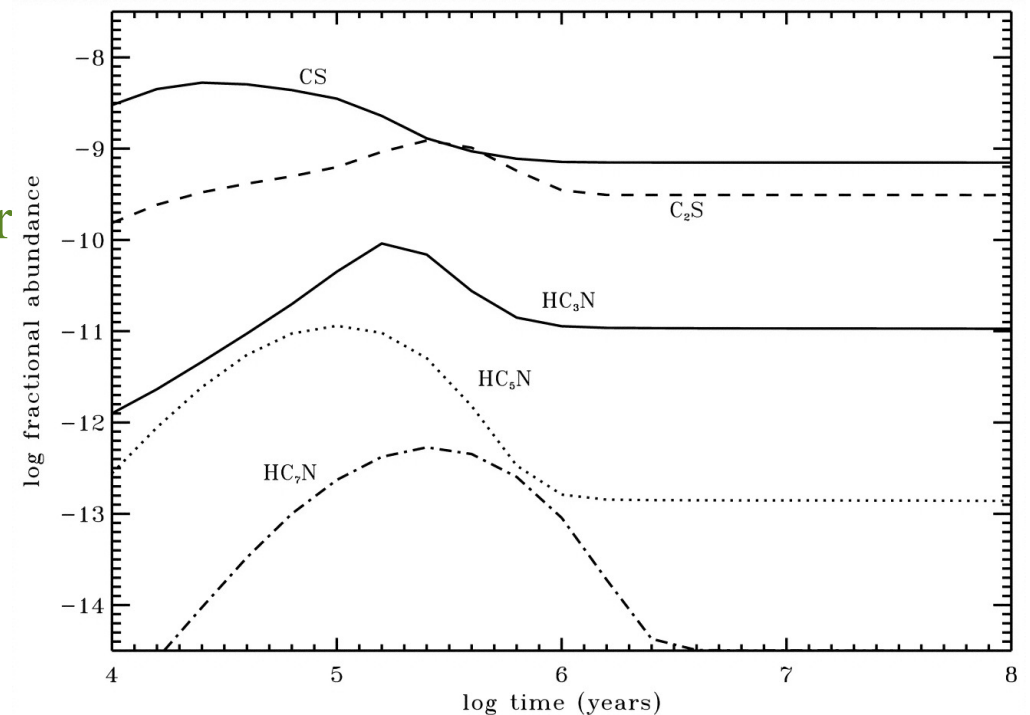
conversion of  $\text{C}^+ \rightarrow \text{C} \rightarrow \text{CO}$  in  $10^6$  yr

Consequence:

early-time / steady-state species

C-chains need C for formation

*Markwick et al 2000*

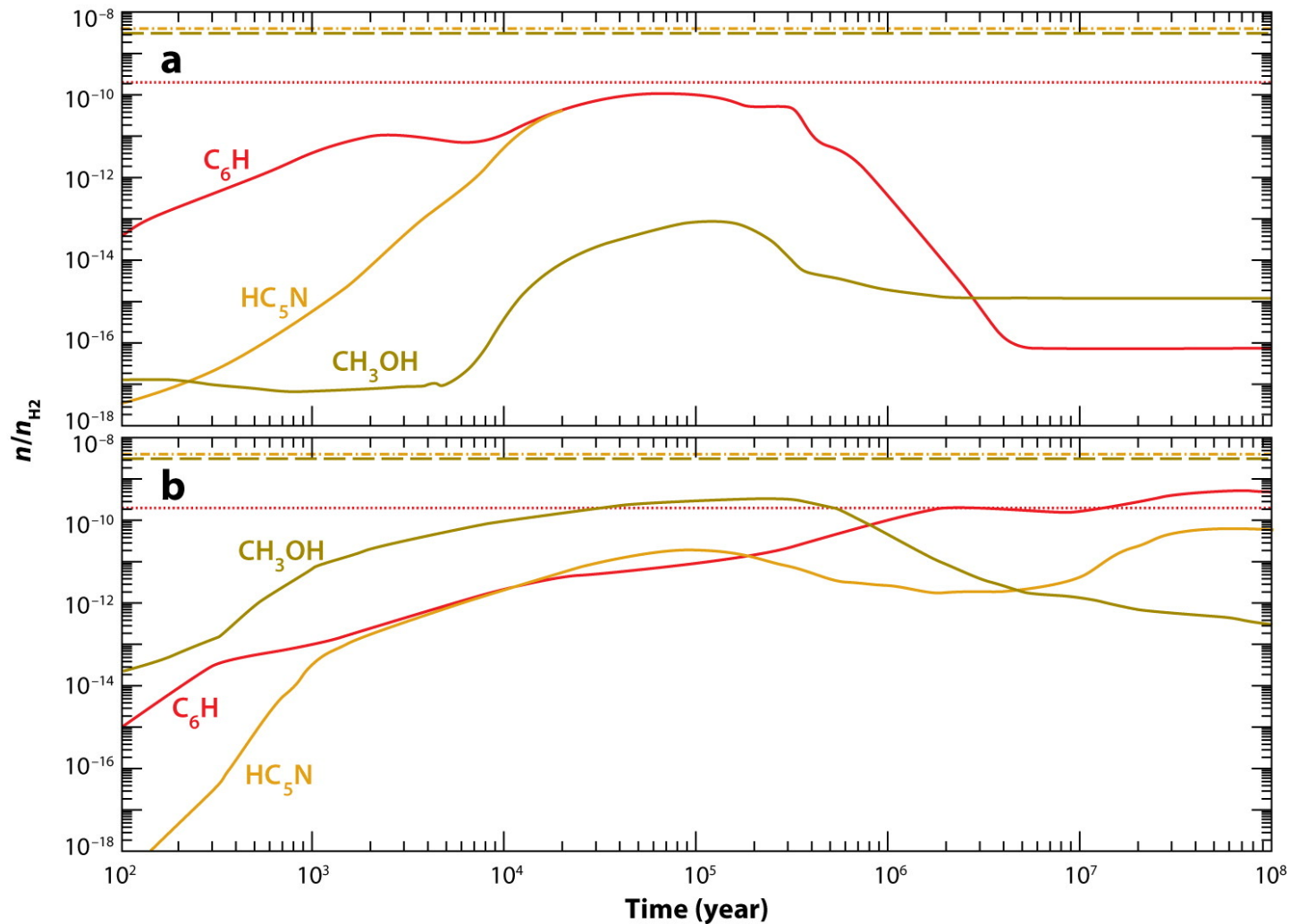


# Methanol in cold clouds

Models with and without grain surface chemistry:

need grains to make observed amount of methanol

*Herbst & van Dishoeck 2009*



# Gas-grain chemistry

Gas depletion timescale:  $2 \times 10^9 y_S^{-1} n_H^{-1} \text{ yr}$

with  $y_S$  = sticking coefficient  $\sim 1$  at 10 K

at  $n_H > 10^4 \text{ cm}^{-3}$ , gas-grain collisions change the chemistry

A few atoms / molecules collide with grain each day

H, He, C, O, CO, N / N<sub>2</sub>

All species but He stick to surface with  $y_S \approx 1$

*(Sulfur & nitrogen reservoirs in dense clouds unknown)*

Light species migrate over surface

tunneling: only H & H<sub>2</sub>

thermal hopping: H, C, N, O

maybe: CH, NH, OH, NH<sub>2</sub>, CH<sub>2</sub>, CH<sub>3</sub>

Reactions occur if barrier low enough ... See next lecture



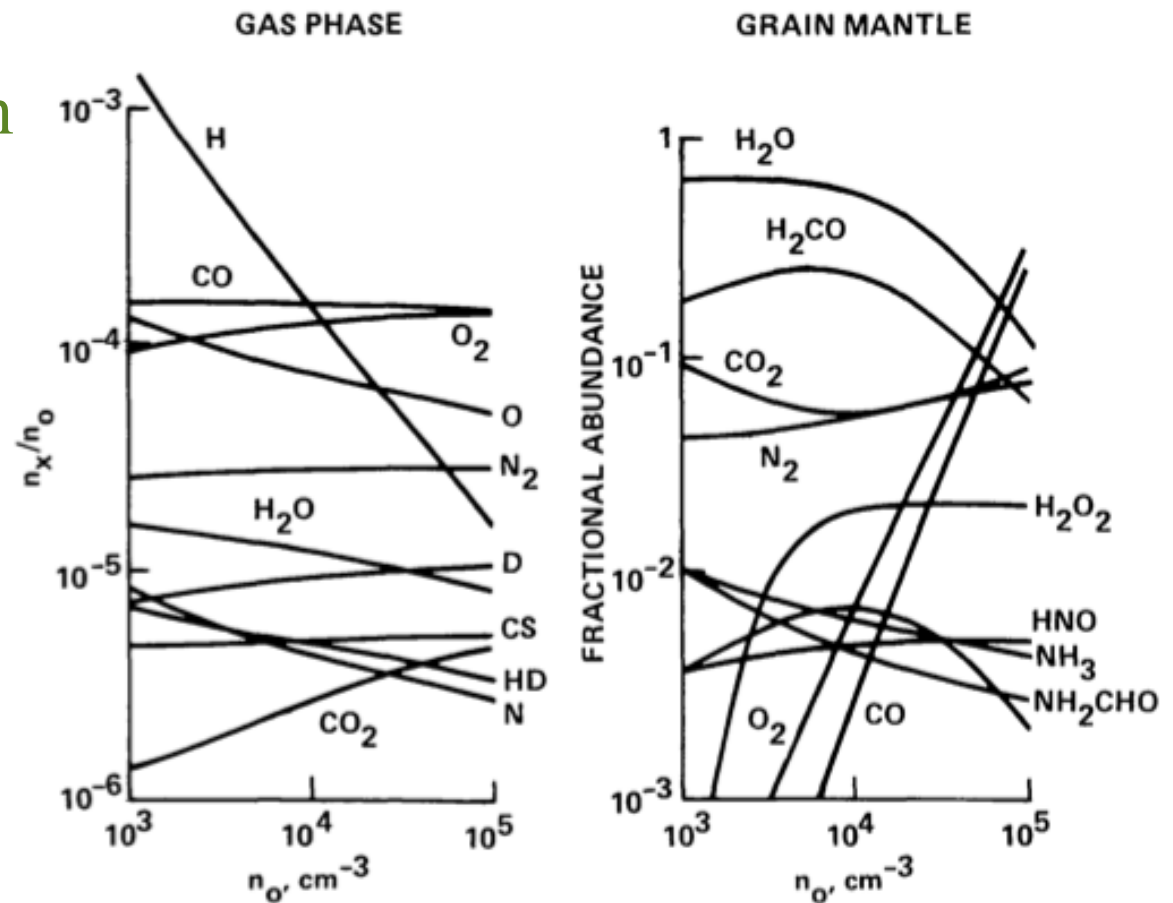
# Interaction between gas phase & grain surface

## Low gas-phase density

larger H abundance (from cosmic-ray ionization):  
mainly *hydrogenation*  
 $\text{H}_2\text{O}$ ,  $\text{CH}_4$ ,  $\text{NH}_3$ , ...

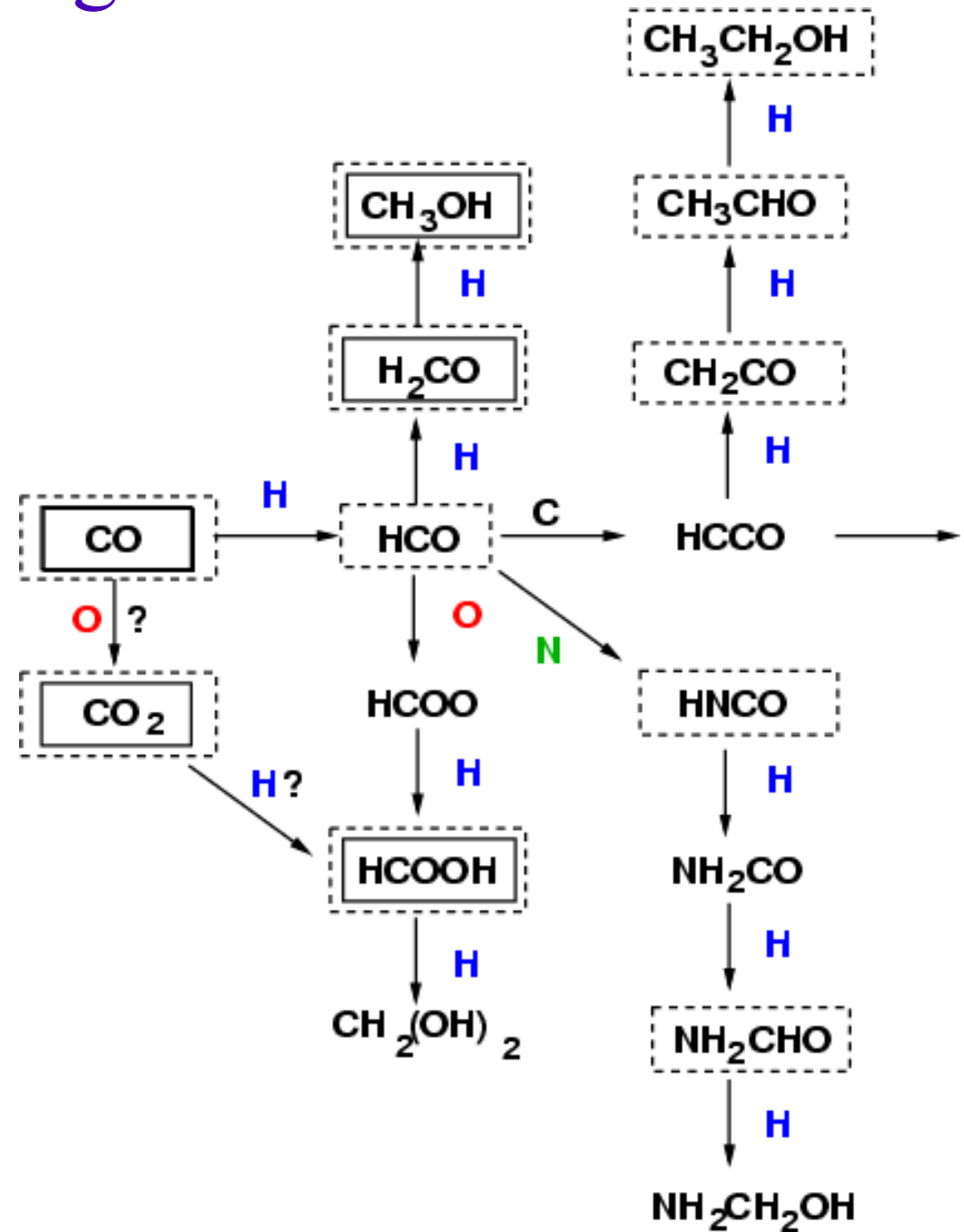
## High gas-phase density

larger O abundance:  
mainly *oxidation*  
 $\text{CO}_2$ ,  $\text{O}_2$ , ...



*Accretion-limited regime:*  
*Tielens 1989*

# Formation of larger molecules



Dashed: detected in gas  
Solid: detected in ice

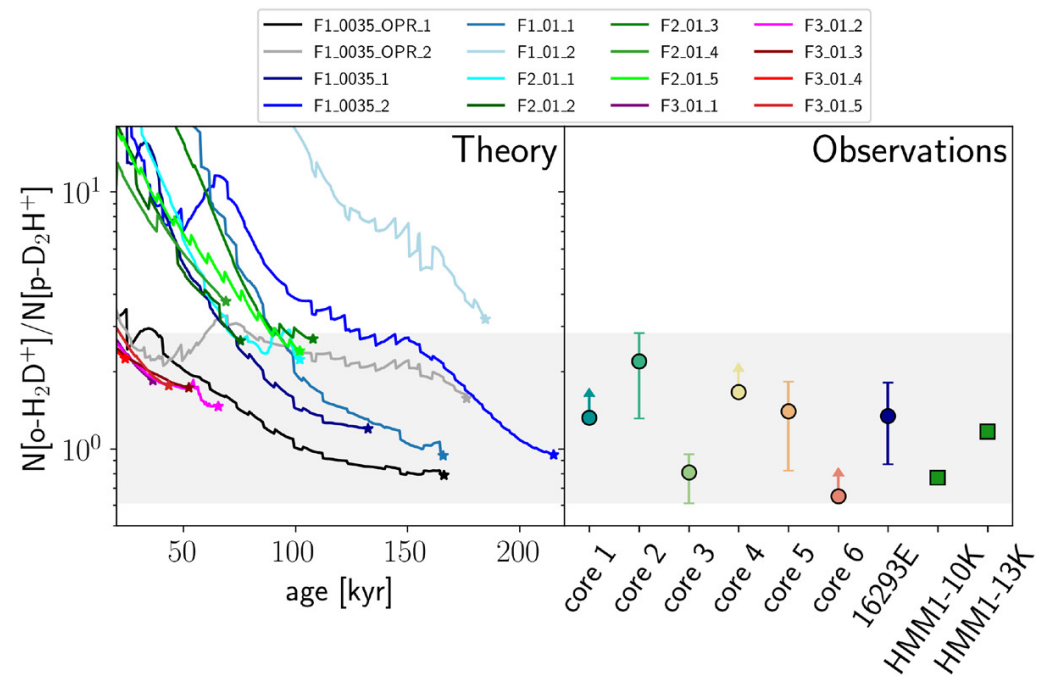
*Bisschop et al 2007*

# The chemical ages of pre-stellar cores

APEX observations of ortho- $\text{H}_2\text{D}^+$  & para- $\text{D}_2\text{H}^+$

6 pre-stellar cores  
in Ophiuchus cloud

Core are chemically young:  
ages  $\sim$  free-fall time  
 $\ll$  ambipolar diffusion



*Brünken et al 2014; Bovino et al 2021*

# Origin of strong deuteration



- **Forward reaction exothermic by 230 K**  
more rapid at low  $T$
- **Reason: zero point vibration of  $H_2D^+$  lower than  $H_3^+$**   
enhanced by nuclear spin statistics (Herbst 1982; Hugo et al 2007)
- **Proton affinity of  $H_2$  low**  
deuteration passed on to other species (CO,  $N_2$ , ...)
- **Freeze-out of CO enhances deuteration**  
main destroyer of  $H_2D^+$  gone

# Chemical structure of dense cores

- **Starless cores: abundant carbon chains**

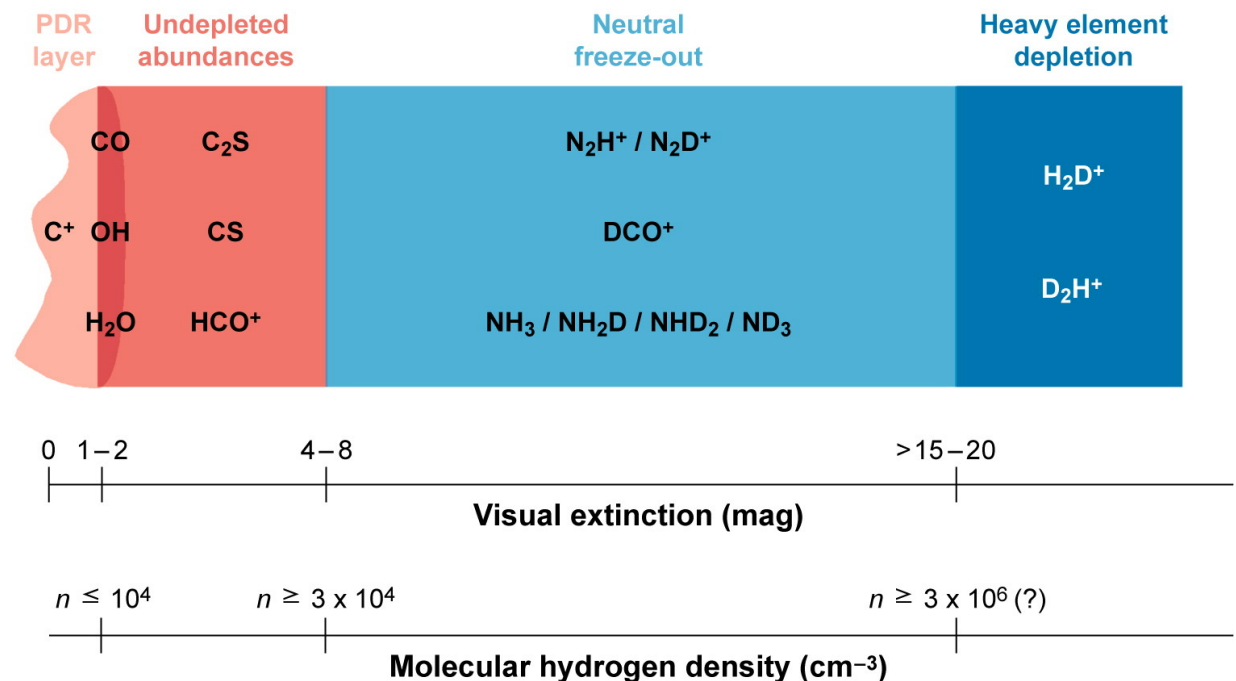
indicates young age  
atomic C available

- **Pre-stellar cores: centrally condensed**

CO freeze-out in center  
N- and D-species enhanced

Grain chemistry essential

Major gas-phase tracers in starless cores



# Summary

- **Chemistry in space is quite unlike on Earth ...**
  - Low densities: Limited by kinetics
  - Low temperatures: Ionization acts as starter
  - Light elements: mainly H with CNO
- **...and varies between environments**
  - Diffuse clouds: Photodissociation (radicals, ions); chemistry depth-dependent
  - Dark clouds: Grain surface processes (organics, ices); chemistry time-dependent

# Exercise: Talking to strangers

## Astronomers: explain to a chemist

- what magnitudes per square arcsecond are
- how dust extinction reddens starlight
- what the stellar and galaxy main sequences are
- how distances to stars and galaxies are measured
- what you mean by ‘metals’, ‘water’, ‘oxygen’, and ‘complex’ molecules

## Chemists: explain to an astronomer

- what the Arrhenius law is
- how Lewis structures work
- how group theory helps to understand molecular symmetry
- what  $\pi$  and  $\sigma$  bonds are, and  $sp^2/sp^3$  hybrids
- what you mean by ‘metals’, ‘water’, ‘oxygen’, and ‘complex’ molecules

# In memoriam Harold Linnartz 1965–2023

- **Career in astrochemistry**

PhD 1994 Nijmegen / Göttingen  
postdoc Bonn & Basel  
researcher VU Amsterdam  
Head of Leiden laboratory

- **Research interests**

DIBs, e.g. EDIBLES survey  
organic molecules  
surface reactions

- **Expertise**

gas-phase / PAH / solid-state  
optical & IR spectra

- **Active member of DAN since inception**





# Interstellar cloud chemistry in context

- **Insight into star & planet formation**  
Inga Kamp, Maryvonne Gerin
- **Molecular structure & interaction**  
Jacques Le Bourlot, Gerrit Groenenboom
- **Basic chemistry at exotic conditions**  
Thanja Lamberts, Alessandra Candian, Valentine Wakelam
- **Link with laboratory technology**  
Sandra Brünken, Sergio Ioppolo

# Enjoy your week!

	Monday 26 August	Tuesday 27 August	Wednesday 28 August	Thursday 29 August	Friday 30 August
9:00		Star & planet formation <i>Inga Kamp</i>	Observational techniques <i>Maryvonne Gerin</i>	Reaction networks <i>Valentine Wakelam</i>	Gas phase laboratory <i>Sandra Brünken</i>
10:00					
10:30		Coffee	Coffee	Coffee	Coffee
11:00		Mesoscopic astrochemistry <i>Alessandra Candian</i>	Laboratory spectroscopy <i>Sandra Brünken / Sergio Ioppolo</i>	Molecular collisions <i>Jacques Le Bourlot</i>	Observational future <i>Maryvonne Gerin</i>
11:45					Laboratory future <i>Sergio Ioppolo</i>
12:00	Registration				Computational future <i>Gerrit Groenenboom</i>
12:30	Sandwich lunch	Buffet lunch	Buffet lunch	Buffet lunch	
13:15					Goodbye & sandwich lunch
13:30	Welcome & logistics				
14:00	Interstellar clouds <i>Floris van der Tak</i>	Molecular structure <i>Thanja Lamberts</i>	Gas phase processes <i>Valentine Wakelam</i>	Laboratory surface physics <i>Sergio Ioppolo</i>	
14:30					
15:00					
15:30	Tea break	Tea break	Tea break	Tea break	
16:00					
16:30	Grain surface processes <i>Thanja Lamberts</i>	Spectroscopy & radiative transfer <i>Jacques Le Bourlot</i>	Exercise session	Social event	
17:00					
17:30	Poster pitches I	Poster pitches II	Poster pitches III		
18:00					
18:30	Buffet dinner	Buffet dinner	Buffet dinner	Barbecue	