Radiative Transfer, Spectroscopy & Collisions - I

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What?

Introduction

Plan PDR Specific Intensity Transfer Space Models Inverse problem

Different problems

Matter properties

Atomic structure

Molecular structure

Carbon Excitation



California Nebulae ⓒ (Astrofalls)



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- Definitions & Problems
- Matter properties



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- Definitions & Problems
- Matter properties
- Spectroscopy:
 - Atomic structure and transitions.
 - Molecular structure.
 - Example: C lines in Orion proplyds.



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 - Molecular structure.
 - Example: C lines in Orion proplyds.
- Radiative transfer II:
 - Interstellar extinction
 - Non LTE situation
 - Some codes



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 - Example: C lines in Orion proplyds.
- Radiative transfer II:
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 - Non LTE situation
 - Some codes
- Collisions:
 - PES.
 - Collision computations.
 - Example: H_3^+ excitation in diffuse clouds.



PDR

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Meudon PDR code.

- ▶ 1D plane parallel steady-state.
- Sophisticated micro-physics

New version 1.7 available (soon).



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Observations: Orion (Bar and d203-506). HST + JWST + ALMA + ...



Haworth et al. (2023) $0.1\,\mathrm{pc}\simeq1\,\mathrm{arc\,mn}$



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Observations: Orion (Bar and d203-506). HST + JWST + ALMA + ...



Haworth et al. (2023) $500au \simeq 1.2 \operatorname{arc\,sec}$



Specific Intensity (aka Spectral Radiance)

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I: Energy transported by the radiation along a "ray".

$$I_{\nu}$$
: J s⁻¹ m⁻² ster⁻¹ Hz⁻¹

$$I_{\lambda}$$
: J s⁻¹ m⁻² ster⁻¹ m⁻¹



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$$I_{\lambda} : \mathrm{J} \, \mathrm{s}^{-1} \, \mathrm{m}^{-2} \, \mathrm{ster}^{-1} \, \mathrm{m}^{-1}$$

Conversion: $I_{\lambda} d\lambda = I_{\nu} d\nu \Rightarrow I_{\lambda} = \frac{c}{\lambda^2} I_{\nu} = \frac{\nu^2}{c} I_{\nu}$





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$$I_{\lambda} : \mathrm{J}\,\mathrm{s}^{-1}\,\mathrm{m}^{-2}\,\mathrm{ster}^{-1}\,\mathrm{m}^{-1}$$

$$J_{\nu} = \frac{1}{4\pi} \int_{4\pi} I_{\nu} (\Omega) \ d\Omega; \quad u_{\nu} = \frac{4\pi}{c} J_{\nu} \left(J \, \mathrm{m}^{-3} \, \mathrm{Hz}^{-1} \right)$$
$$F_{\nu} = \int_{2\pi} I_{\nu} (\Omega) \, \cos \theta \, d\Omega; \quad F_{\nu} = \pi \, I_{\nu} \left(I_{\nu} = Cte \right)$$



Transfer



Transfer



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"OUT" = "IN" + "injected" - "removed":

$$\frac{dI}{ds} = -\kappa \, I + \eta$$

With scattering:

$$\frac{dI_{\Omega}}{ds} = -(\kappa + \sigma) I_{\Omega} + \eta + \sigma \int_{4\pi} p(\Omega, \Omega') I_{\Omega'} d\Omega'$$



Space (reality vs. models)

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Building a model:





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Building a model:





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Building a model:





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Building a model:

World \rightarrow Physics \rightarrow Maths \rightarrow Numeric \rightarrow Data



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Building a model:



Start from the real problem.



Carbon Excitation

Models

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Start from the real problem. Check back if the "solution" tells something relevant on this problem.

Physics

Maths

Numeric

Building a model:

Data



Carbon Excitation

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Building a model:



Start from the real problem. Check back if the "solution" tells something relevant on this problem.

Consequence: There are many radiative transfer theories.



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Variables

Observables

Model: 1 to 1 application (most of the time).



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Variables

Observables

Explore sensitivity analysis!



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What you really need.



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Variables

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What you get.



Different problems

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Divide & conquer

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Choose a geometry

- $3D \Rightarrow I(x, y, z, \theta, \phi)$: Use Monte-Carlo (see e.g. LIME (Brinch & Hogerheijde))
- 1D (maybe 2+1D) \Rightarrow Check the physics



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- Is scattering significant?
 - Yes: all directions coupled. No: "follow that ray".



Dust Absorption and Scattering





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 - Yes: all directions coupled. No: "follow that ray".
- Is line transfer significant?
 - Yes: detailed balance is required.



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- Is line transfer significant?
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	No scattering	Scattering
No lines	Easy	Doable
Lines	Hard	



No scattering - No lines





No scattering - No lines

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$$\frac{dI(s,\theta)}{ds} = -\kappa(x) I(s,\theta) + \eta(x)$$

Change of variables:

$$\mu = \cos \theta; \quad ds = \frac{dx}{\mu}; \quad d\tau = \kappa \, dx$$

$$u \frac{dI}{d\tau} = -I + \frac{\eta}{\kappa} = -I + S$$

Constant properties (S = Cte):

$$I(\tau,\mu) = I_0(\mu) \exp\left(-\frac{\tau}{\mu}\right) + S\left(1 - \exp\left(-\frac{\tau}{\mu}\right)\right)$$


No scattering - No lines





Formal solution to the transfer equation

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If κ and η depend on position:

$$\tau\left(s\right) = \int_{0}^{s} \kappa\left(t\right) \, dt$$

$$I(\tau) = I_0 \exp(-\tau) + \int_0^{\tau} S(t) \exp(-(\tau - t)) dt$$



Formal solution to the transfer equation

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$$I(\tau) = I_0 \exp(-\tau) + \int_0^{\tau} S(t) \exp(-(\tau - t)) dt$$

Check:

$$\frac{I(\tau)}{d\tau} = -I_0 \exp(-\tau) + S(\tau) \exp(-(\tau - \tau))$$
$$-\int_0^\tau S(t) \exp(-(\tau - t)) dt$$
$$\frac{I(\tau)}{d\tau} = -I(\tau) + S(\tau)$$

Formal solution if κ and η are known (or given...).



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What if
$$(\mu = \cos \theta, d\tau = (\kappa + \sigma) ds)$$
:

$$\mu \frac{dI(\tau, \mu)}{d\tau} = -I(\tau, \mu) + \frac{\omega}{2} \int_{-1}^{+1} p(\mu, \mu') I(\tau, \mu') d\mu' + S(\tau)$$

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In 1D: use Legendre expansion (see Roberge (1983) for an introduction).

$$I(\tau, \mu) = \sum_{l=0}^{\infty} (2l+1) f_l(\tau) P_l(\mu)$$

 $p(\mu, \mu') = \sum_{l=0}^{\infty} (2l+1) \sigma_l P_l(\mu) P_l(\mu')$



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$$p(\mu, \mu') = \sum_{l=0}^{\infty} (2l+1) \sigma_l P_l(\mu) P_l(\mu')$$

A bit of math magic leads to $(\forall l)$:

 $l f'_{l-1} + (l+1) f'_{l+1} = (2l+1) (1 - \omega \sigma_l) f_l + s \delta_{l0}$



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Truncate at L finite \Rightarrow set of linear equation with non-constant coefficients and boundary conditions at both edges:

 $\mathbf{A}\mathbf{f}^{\prime}\left(\tau\right)=\mathbf{B}\mathbf{f}\left(\tau\right)+\mathbf{g}\left(\tau\right)$

 ${\bf A}$ and ${\bf B}$ diagonally dominant matrices.



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Truncate at L finite \Rightarrow set of linear equation with non-constant coefficients and boundary conditions at both edges:

 $\mathbf{A}\mathbf{f}'(\tau) = \mathbf{B}\mathbf{f}(\tau) + \mathbf{g}(\tau)$

- ${\bf A}$ and ${\bf B}$ diagonally dominant matrices.
- Solved by diagonalization.
 - Formal solution used for each f_l .
 - Requires iterations.
 - Very sensitive to numerical errors (precision).

 \Rightarrow Quite tricky, but doable.

May be extended to include lines.



General case strategy

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Possible strategy:

- 1. Assume values of $\kappa(x)$ and $\eta(x)$ from educated guesses.
- 2. Compute I(x) by the formal solution.
- 3. Update κ and η if they depend on I
- 4. Iterate...



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- 4. Iterate...

Problem: Usually, 1 iteration propagates the solution by 1 photon mean free path $\sim 1/\kappa.$

 \Rightarrow needs special methods (ALI: "Accelerated Lambda Iteration").



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- 4. Iterate...
- Problem: Usually, 1 iteration propagates the solution by 1 photon mean free path $\sim 1/\kappa.$
- \Rightarrow needs special methods (ALI: "Accelerated Lambda Iteration").

Requirement:

Finding κ and η .



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Einstein coefficients

Detailed balance

Critical density

Excitation diagram

Mean radiation field

Lorentz profile

Gaussian profile

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Transmission curves for the NIRISS filters aboard the JWST.



Continuous processes

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Introd	luction
	action

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Many possibilities:

. . .

- Dust properties: absorption is constant, emission depends on temperature (radiative balance)
- Free-free emission and absorption, electron scattering, bremsstrahlung,
- Bound-free continuous absorption by species (Photo-destruction cross-sections.)



Continuous processes

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Many possibilities:

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- Dust properties: absorption is constant, emission depends on temperature (radiative balance)
- Free-free emission and absorption, electron scattering, bremsstrahlung,
- Bound-free continuous absorption by species (Photo-destruction cross-sections.)
- Transfer usually converges rapidly. Requires access to many data bases for micro-physics processes. E.g.:
 - Dust: Bruce Draine's web site.
 - Photo-destruction: Ewine van Dishoeck web site.
 - Stellar models. E.g. CESAM2k20.
- Check detailed balance!



Lines

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Absorption and emission coefficients depend on species level populations $(n_u \text{ and } n_l)$:

$$\kappa_{lu} = \frac{h c}{4\pi \lambda} \left(B_{lu} n_l - B_{ul} n_u \right) \phi_{\lambda}$$

$$\eta_{ul} = \frac{h c}{4\pi \lambda} A_{ul} n_u \phi_\lambda$$

 A_{ul} , B_{ul} and B_{lu} : Einstein coefficients. ϕ_{λ} : Line profile - Gaussian or Voigt (see later). n_u (resp. n_l): Population of "upper" (resp. "lower") level of the transition.



Einstein coefficients

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 A_{ul} : Probability of spontaneous transition, in s⁻¹. \Rightarrow Find in relevant data bases (e.g. NIST) B_{ul} and B_{ul} : Induced transitions. Depend on the radiati

 B_{ul} and B_{lu} : Induced transitions. Depend on the radiative field representation. For I_{λ} and specific intensity:

$$A_{ul} = \frac{2hc^2}{\lambda^5} B_{ul}; \quad g_u B_{ul} = g_l B_{lu}$$

Conversion $\lambda \leftrightarrow \nu$ and/or $I_{\lambda} \leftrightarrow u_{\lambda}$ is a nightmare...



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$$A_{ul} = \frac{2 h c^2}{\lambda^5} B_{ul}; \quad g_u B_{ul} = g_l B_{lu}$$

Conversion $\lambda \leftrightarrow \nu$ and/or $I_{\lambda} \leftrightarrow u_{\lambda}$ is a nightmare...

Compute populations



Detailed balance (2 levels)





Detailed balance:

$$n_{l} \left(B_{lu} \,\bar{J}_{lu} + k_{lu}^{X} \,n_{X} + D_{l} \right) = n_{u} \left(A_{ul} + B_{ul} \,\bar{J}_{ul} + k_{ul}^{X} \,n_{X} \right) + F_{l}$$
$$n_{u} \left(A_{ul} + B_{ul} \,\bar{J}_{ul} + k_{ul}^{X} \,n_{X} + D_{u} \right) = n_{l} \left(B_{lu} \,\bar{J}_{lu} + k_{lu}^{X} \,n_{X} \right) + F_{u}$$



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Collisional rates linked by detailed balance:

$$g_l k_{lu} = g_u k_{ul} \exp\left(-\frac{E_{ul}}{k T}\right)$$



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Collisional rates linked by detailed balance:

$$g_l \, k_{lu} = g_u \, k_{ul} \, \exp\left(-\frac{E_{ul}}{k \, T}\right)$$

- Sum equations: $n_l D_l + n_u D_u = F_l + F_u$. But $D_l \neq D_u$ and $F_l \neq F_u$ in general!
 - System is well behaved. $n_l + n_u = n_{tot}$ not needed. \Rightarrow Check on computation!



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 - System is well behaved. $n_l + n_u = n_{tot}$ not needed. \Rightarrow Check on computation!
- Neglecting induced transitions and Formation/Destruction:

$$n_l k_{lu} n_X = n_u (A_{ul} + k_{ul} n_X)$$



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 - System is well behaved. $n_l + n_u = n_{tot}$ not needed. \Rightarrow Check on computation!

Neglecting induced transitions and Formation/Destruction:

$$n_l k_{lu} n_X = n_u \left(A_{ul} + k_{ul} n_X \right)$$

$$\frac{n_u}{n_l} = \frac{1}{1 + \frac{A_{ul}}{k_{ul}^X n_X}} \frac{g_u}{g_l} \exp\left(-\frac{E_{ul}}{kT}\right)$$



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$$\frac{n_u}{n_l} = \frac{1}{1 + \frac{A_{ul}}{k_{ul} n_X}} \frac{g_u}{g_l} \exp\left(-\frac{E_{ul}}{k T}\right)$$

High density: $k_{ul} n_X \gg A_{ul} \Rightarrow n_u$ and n_l given according to a Boltzmann distribution at temperature T.



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$$\frac{n_u}{n_l} = \frac{1}{1 + \frac{A_{ul}}{k_{ul} n_X}} \frac{g_u}{g_l} \exp\left(-\frac{E_{ul}}{kT}\right)$$

High density: $k_{ul} n_X \gg A_{ul} \Rightarrow n_u$ and n_l given according to a Boltzmann distribution at temperature T.

Low density: Upper level is radiatively depopulated,

with correction factor
$$1/\left(1+\frac{A_{ul}}{k_{ul} n_X}\right)$$
.



Critical density

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High density: $k_{ul} n_X \gg A_{ul} \Rightarrow n_u$ and n_l given according to a Boltzmann distribution at temperature T.

Low density: Upper level is radiatively depopulated, with correction factor $1/\left(1 + \frac{A_{ul}}{k_{ul} n_X}\right)$.

Critical density: n_c such that:

$$\frac{A_{ul}}{k_{ul} n_c} = 1; \quad n_c = \frac{A_{ul}}{k_{ul}}$$

I n_c depends on the line considered!



Critical density - C^+



Example: C⁺ - $A_{ul} = 2.29 \, 10^{-6} \, \text{s}^{-1}$, $\frac{E_{ul}}{k} = 91.25 \, \text{K}$, $k_{ul} \simeq 8.9 \, 10^{-16} \, \text{m}^3 \, \text{s}^{-1}$ with H (dominates e^-).



Excitation diagram

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Pure Boltzmann distribution:

$$n_j \propto g_j \exp\left(-\frac{E_j}{kT}\right); \quad \ln\left(\frac{n_j}{g_j}\right) = -\frac{1}{T}\frac{E_j}{k} + Cte$$



Excitation diagram

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10000

E_J (K)

15000



5000

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28

0

20000



Mean radiation field ($B_{ul} \bar{J}_{ul}$)

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 \overline{J}_{ul} is the local mean radiation field. That is:

$$\bar{J}_{ul} = \int_{-\infty}^{+\infty} \left(\frac{1}{4\pi} \int_{4\pi} I_{\nu} \left(\Omega\right) \, d\Omega\right) \, \phi_{\nu} \, d\nu$$



Mean radiation field $(B_{ul} \bar{J}_{ul})$

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• Line profile ϕ_{ν} . Such that:

$$\int_{-\infty}^{+\infty} \phi_{\nu} \, d\nu = 1$$



Mean radiation field ($B_{ul} \bar{J}_{ul}$)

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Line profile ϕ_{ν} . Such that:

$$\int_{-\infty}^{+\infty} \phi_{\nu} \, d\nu = 1$$

Broadening:

- Thermal \Rightarrow Gaussian profile.
- Turbulent \Rightarrow Gaussian profile.
- Collisional \Rightarrow Lorentz profile.
- Natural \Rightarrow Lorentz profile.
- ♦ Combined ⇒ Convolution of Gaussian and Lorentz
 ⇒ Voigt profile.



Lorentz profile

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$$\phi_L(\nu) = \frac{1}{\pi} \frac{\Gamma/4\pi}{\left(\nu - \nu_{ul}\right)^2 + \left(\Gamma/4\pi\right)^2}$$

 Γ : "damping constant": sum of all spontaneous transition probabilities (inverse radiative lifetime). Full width at half maximum (FWHM):

$$\gamma_L = \frac{\Gamma}{2\pi}$$

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Gaussian profile

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Combine thermal and micro-turbulent velocity v_t
 Doppler width:

$$\Delta \nu_D(s) = \frac{\nu_{ul}}{c} \sqrt{\frac{2kT(s)}{\mu} + v_t^2(s)} = \frac{\nu_{ul}}{c} v_T(s)$$

Profile:

$$\phi_D(\nu) = \frac{1}{\sqrt{\pi} \Delta \nu_D} \exp\left(-\left(\frac{\nu - \nu_{ul}}{\Delta \nu_D}\right)^2\right)$$
$$= \frac{1}{\sqrt{\pi} \nu_{ul}} \frac{c}{v_T(s)} \exp\left(-\left(\frac{\nu - \nu_{ul}}{\nu_{ul}} \frac{c}{v_T(s)}\right)^2\right)$$

$$\gamma_D = 2\sqrt{\log 2}\,\Delta\nu_D$$



Voigt profile

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Convolution of Gaussian and Lorentz:

$$\phi_V(\nu) = \int_{-\infty}^{+\infty} \phi_L(\nu_{ul} + \nu - \nu') \phi_D(\nu') d\nu'$$



Voigt profile

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Convolution of Gaussian and Lorentz:

$$\phi_V(\nu) = \int_{-\infty}^{+\infty} \phi_L(\nu_{ul} + \nu - \nu') \phi_D(\nu') d\nu'$$

Let
$$a = \frac{1}{\Delta \nu_D} \frac{\Gamma}{4\pi}$$
, $x = \frac{\nu - \nu_{ul}}{\nu_{ul}}$, then:
 $\phi_V(\nu) = \frac{1}{\Delta \nu_D} \frac{1}{\sqrt{\pi}} H(a, x)$

with:

$$H(a, x) = \frac{a}{\pi} \int_{-\infty}^{+\infty} \frac{\exp(-y^2)}{(x-y)^2 + a^2} \, dy$$



Voigt profile






Atomic structure and transitions





Radiative transitions

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Radiative transitions

Electrons

arrangement

Total Angular momentum

Terms C $(1s^2 2s^2 2p^2)$

Selection rules

Molecular structure

Carbon Excitation

Governed by selection rules.

Based on quantum mechanics constraints.

Transition probability: $\langle \Psi_i | \mu | \Psi_f \rangle$: an integral...

Is 0 if some symmetry properties are not met.

• Depend mostly on total angular momentum of initial (Ψ_i) and final (Ψ_f) state.







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Electrons are characterized by quantum numbers:

- *n*: number of nodes in the wave function. n > 0
- l: quantum of angular momentum. l < n
- m_z : projection of angular momentum on O'_z . $-l \leqslant m_z \leqslant +l$
- s: spin (intrinsic angular momentum). $s=\pm rac{1}{2}$



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- Electrons are fermions:
 - \Rightarrow No 2 e^- can have the same quantum numbers.



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 - (sub-)shell: same n and l. Number of e^- : at most 2 (2l + 1)





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 - \Rightarrow No 2 e^- can have the same quantum numbers.
 - (sub-)shell: same n and l. Number of e^- : at most 2 (2l + 1)Standard notation for l: s, p, d, f...
- Example: 6 e^- (C, N⁺, O⁺⁺,...):

 $1s^2 \, 2s^2 \, 2p^2$



Total Angular momentum

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Carbon Excitation

1s and 2s shells are full: contribute 0. 2p shell: 6 possibilities for each e^- and only 2 e^- . Same n and same l, so they must have different m_z (3 possibilities) and s (2 possibilities). Number of pairs: $\frac{6 \times 5}{2} = 15$Because of Pauli principle.



Total Angular momentum



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1s and 2s shells are full: contribute 0. 2p shell: 6 possibilities for each e^- and only $2e^-$. Same n and same l, so they must have different m_z (3 possibilities) and s (2 possibilities). Number of pairs: $\frac{6 \times 5}{2} = 15$Because of Pauli principle. Individual e^- angular momentum and spin momentum are combined to build the total angular momentum:

- Angular momentum: L
- Spin momentum: S
- Total momentum: J, with $|L S| \leq J \leq |L + S|$



Total Angular momentum

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Total Angular momentum

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Carbon Excitation

1s and 2s shells are full: contribute 0.
2p shell: 6 possibilities for each e⁻ and only 2 e⁻.
Same n and same l, so they must have different m_z (3 possibilities) and s (2 possibilities).
Number of pairs: 6×5/2 = 15. ...Because of Pauli principle.
Individual e⁻ angular momentum and spin momentum are combined to build the total angular momentum:
Angular momentum: L
Spin momentum: S

• Total momentum: J, with $|L - S| \leq J \leq |L + S|$

Terms (with L = S, P, D, ...):

 $^{2S+1}L_J$



Terms for C

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- Terms C $(1s^2 2s^2 2p^2)$
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Scan possibilities for m_z and s, for fixed n = 2 and l = 1: Case 1: Both e^- have $m_z = 1$.

♦ Then, L = 2, which implies S = 0 and J = 2.
♦ Only one possibility: ¹D₂, g = 2J + 1 = (2S + 1) (2L + 1) = 5.



Terms for ${\rm C}$

	1.1		
ntro	duo	ctio	n

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- ♦ Then, L = 2, which implies S = 0 and J = 2.
 ♦ Only one possibility: ¹D₂, g = 2J + 1 = (2S + 1) (2L + 1) = 5.
 Case 2: Both e⁻ have s = +¹/₂.
 - Then, S = 1. m_z must differ, so $L \ge 1$. • But L = 2 is not possible, so L = 1.
- Hence it is ${}^{3}P$, g = 9. 3 possibilities for J: 0, 1 or 2.
- ${}^{3}P_{0}, g = 1, {}^{3}P_{1}, g = 3, {}^{3}P_{2}, g = 5 \text{ and } 1 + 3 + 5 = 9.$



Terms for ${\rm C}$

	ntrod	luction
	ILLOU	
-		

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Scan possibilities for m_z and s, for fixed n = 2 and l = 1: Case 1: Both e^- have $m_z = 1$.

- Then, L = 2, which implies S = 0 and J = 2.
 Only one possibility: ¹D₂, g = 2J + 1 = (2S + 1) (2L + 1) = 5.
 Case 2: Both e⁻ have s = +¹/₂.
 Then S = 1 m must differ so L ≥ 1.
 - Then, S = 1. m_z must differ, so $L \ge 1$.
 - But L = 2 is not possible, so L = 1.
 - Hence it is ${}^{3}P$, g = 9. 3 possibilities for J: 0, 1 or 2.
 - ${}^{3}P_{0}, g = 1, {}^{3}P_{1}, g = 3, {}^{3}P_{2}, g = 5 \text{ and } 1 + 3 + 5 = 9.$
- Case 3: Already 5 + 9 = 14 cases accounted for.
 - Only 1 left. Hence S = 0 and L = 0.
 - So the last one is 1S_0 .



Terms for ${\rm C}$

	ntrod	luction
	ILLOU	
-		

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Scan possibilities for m_z and s, for fixed n = 2 and l = 1: Case 1: Both e^- have $m_z = 1$.

- Then, L = 2, which implies S = 0 and J = 2.
 Only one possibility: ¹D₂, g = 2J + 1 = (2S + 1) (2L + 1) = 5.
 Case 2: Both e⁻ have s = +¹/₂.
 Then, S = 1. m_z must differ, so L ≥ 1.
 But L = 2 is not possible, so L = 1.
 Hence it is ³P, g = 9. 3 possibilities for J: 0, 1 or 2.
 ³P₀, g = 1, ³P₁, g = 3, ³P₂, g = 5 and 1 + 3 + 5 = 9.
- Case 3: Already 5 + 9 = 14 cases accounted for.
 - Only 1 left. Hence S = 0 and L = 0.
 - So the last one is 1S_0 .

Energies: Solve Schrödinger.

C $(1s^2 2s^2 2p^2)$





Selection rules

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Selection rules

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Carbon Excitation

In $\langle \Psi_i | \mu | \Psi_f \rangle$, μ is expended on spherical harmonics $Y_l^m (\theta, \phi)$.

Expansion gives electric and magnetic multipoles contribution E_1 , E_2 , ... M_1 , M_2 ...

Each term in the expansion has its symmetry and gives selection rules.



Selection rules

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Carbon Excitation

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Expansion gives electric and magnetic multipoles contribution E_1 , E_2 , ... M_1 , M_2 ...

Each term in the expansion has its symmetry and gives selection rules.

Main ones:

 ♦ If dipole transition (E₁, M₁) are possible (P, Q and R branches):

 $\Delta J = 0, \pm 1; \quad 0 \nleftrightarrow 0$

 $P:\,J\rightarrow J+1,\,Q:\,J\rightarrow J,\,R:\,J\rightarrow J-1$

• For $J' \to J$ " transition: P(J"), Q(J"), R(J").



Selection rules

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Carbon Excitation

In $\langle \Psi_i | \mu | \Psi_f \rangle$, μ is expended on spherical harmonics $Y_l^m(\theta, \phi)$.

Expansion gives electric and magnetic multipoles contribution E_1 , E_2 , ... M_1 , M_2 ...

Each term in the expansion has its symmetry and gives selection rules.

Main ones:

• If not: E_2 , M_2 (O and S branches):

$$\Delta J = 0, \pm 1, \pm 2; \quad 0 \nleftrightarrow 0, 1; \quad \frac{1}{2} \nleftrightarrow \frac{1}{2}$$

• Strongest: E_1 , then $E_2 \sim M_1$, then $E_3 \sim M_2$...



Molecular structure

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 \bigodot Physics World



Born-Oppenheimer approximation

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- Separate electrons from nucleons.
- Electrons adapt adiabatically to nucleons positions.
- Nucleon motions:

- Vibration \Rightarrow No selection rules.
- Rotation \Rightarrow Angular momentum selection rules.
- Branches: v' v'' P(J''), v' v'' Q(J''), v' v'' R(J'').
- Database: HITRAN.



Born-Oppenheimer approximation

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- Database: HITRAN.

Diagnostics possible from far UV to Radio



© JWST & ALMA



Molecular hydrogen





Main electronic states: $X^{1}\Sigma_{g}^{+}$, $B^{1}\Sigma_{u}^{+}$ (Lyman transitions), $C^{1}\Pi_{u}$ (Werner transitions)

 Infrared spectrum: Ro-vibrational quadrupolar transitions within X¹∑⁺_g.
 Ultraviolet spectrum: Electronic transitions ⇒ lead to dissociation (~10%).



Molecular hydrogen

H(1s) + H(3l)

H(1s) + H(2l)

H(1s) + H(1s)

12

Lyman limit

R (a.u.)

10

8

←



Main electronic states: $X^{1}\Sigma_{a}^{+}$, $B^{1}\Sigma_{u}^{+}$ (Lyman transitions), $C^{1}\Pi_{u}$ (Werner transitions) Infrared spectrum:

Ro-vibrational quadrupolar transitions within $X^{1}\Sigma_{a}^{+}$.

Ultraviolet spectrum: Electronic transitions \Rightarrow lead to dissociation $(\sim 10\%).$

Dominates edge of PDR.

H2

C+

C

CO

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Molecular region



Absorption from H_2

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- UV pumping followed by radiative decay:
 - 10% dissociation.
 - 90% populates high v (up to v = 14), but low J.
 - $\lambda_{emis} > \lambda_{abs}$



Absorption from ${\rm H}_2$

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- UV pumping followed by radiative decay:
 - 10% dissociation.
 - 90% populates high v (up to v = 14), but low J.
 - $\lambda_{emis} > \lambda_{abs}$
- Dissociating photons disappear close to the cloud surface.
 - \Rightarrow "Self-Shielding": H₂ protects H₂.
 - Less UV photons for heating, ionization, dissociation...
- Example: d203-506 (constant density model):

$$\begin{array}{c|c} n_{\rm H} \ ({\rm m}^{-3}) & G_0 & \zeta \ ({\rm s}^{-1}) \\ 10^{13} & 2.3 \ 10^4 & 10^{-16} \end{array}$$























 $A_{\rm V} = 0.1$, Absorption from 302 levels + fluorescence.

















Carbon Excitation







Carbon Excitation





Carbon Excitation in d203-506

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Carbon Excitation Spectrum

Simple model NIST Database Carbon structure Simple model Carbon structure C in d203-506 No O lines?

[CI] ${}^{1}D_{2}-{}^{3}P_{2}$ 0.9853 μ m



d203-506 in Orion bar - JWST - Goicoechea et al. (2024)



Spectrum d203-506








Simple model



Meudon PDR code - Standard parameters



NIST Database

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Carbon Excitation Spectrum

Simple model

NIST Database

Carbon structure Simple model Carbon structure C in d203-506 No O lines?

Many reliable info at NIST:

- lons and atoms Levels.
- lons and atoms Lines.
- Good for ions and atoms No molecule...



Carbon structure

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Simple model



Meudon PDR code - Standard parameters



Simple model + Recombination





Carbon structure (bis)

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C in d203-506 No O lines?

















$\rm C$ in d203-506

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Matter properties	Iransition		corrected	(pump)	(recomb)
Atomic structure		(μm)	$({\rm erg}{\rm cm}^{-2}{\rm s}^{-1}{\rm ster}^{-1})$		
Molecular structure	${}^{3}P_{1} - {}^{3}P_{0}$	609.14	1.6210^{-6}	1.6010^{-6}	1.6010^{-6}
Carbon Excitation	${}^{1}D_{2} - {}^{3}P_{1}$	0.9827	1.9410^{-4}	1.1310^{-4}	2.0210^{-6}
Simple model	${}^{1}D_{2} - {}^{3}P_{2}$	0.9853	6.5710^{-4}	3.3810^{-4}	9.1710^{-6}
NIST Database	$(3p)^{3}D - {}^{3}P^{o}$	1.069	3.2310^{-4}	2.3410^{-4}	1.4010^{-6}
Simple model	$(3d)^{3}F^{o} - {}^{3}D$	1 1 7 6	3.0810^{-5}	4.7810^{-5}	$6.40 10^{-7}$
Carbon structure	(Ju) T = D	1.170	0.0010	+.1010	0.4010
C in d203-506	$(4s)^{3}P^{3} - (3p)^{3}S$	1.355	4.5510^{-5}	3.30 10 8	8.34 10
No U lines?					

C lines come from UV pumping, not recombination.



No O lines?



